

**PCT**WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification <sup>6</sup> :</b> <b>C07D 215/48, 215/38, C07C 309/51,</b> <b>A61K 31/47, 31/185</b>	<b>A1</b>	<b>(11) International Publication Number:</b> <b>WO 96/25399</b> <b>(43) International Publication Date:</b> 22 August 1996 (22.08.96)
<b>(21) International Application Number:</b> PCT/US96/02134 <b>(22) International Filing Date:</b> 16 February 1996 (16.02.96) <b>(30) Priority Data:</b> 08/390,057 17 February 1995 (17.02.95) US <b>(71) Applicants:</b> THE UNITED STATES OF AMERICA, represented by THE SECRETARY, DEPARTMENT OF HEALTH AND HUMAN SERVICES National Institutes of Health [US/US]; Office of Technology Transfer, Box OTT, Bethesda, MD 20892-9902 (US). GEORGIA TECH RESEARCH CORPORATION [US/US]; 400 Tenth Street, Atlanta, GA 30332-0415 (US). <b>(72) Inventors:</b> HAUGWITZ, Rudiger, D.; 5915 Greenlawn Drive, Bethesda, MD 20814 (US). ZALKOW, Leon; 170 Zebelin Road, Atlanta, GA 30342 (US). GRUSZECKA-KOWALIK, Ewa; 500 Northside Circle, N.W. #E-6, Atlanta, GA 30309 (US). BURGESS, Edward; 22 Lakeshore Drive, Marietta, GA 30067 (US). <b>(74) Agents:</b> KILYK, John, Jr. et al.; Leydig, Voit & Mayer, Ltd., Suite 4900, Two Prudential Plaza, 180 North Stetson, Chicago, IL 60601-6780 (US).	<b>(81) Designated States:</b> AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).  <b>Published</b> <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>	
<b>(54) Title:</b> NOVEL AROYLANILINE DERIVATIVES, COMPOSITIONS AND USES THEREOF IN THE TREATMENT OF VIRAL INFECTIONS  <b>(57) Abstract</b>  Aroylaniline derivatives which exhibit anti-retroviral activity, pharmaceutical compositions containing such aroylaniline derivatives, and methods for treating a retroviral-infected host comprising administering an antiviral effective amount of an aroylaniline derivative to a host.		

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AM	Armenia	GB	United Kingdom	MW	Malawi
AT	Austria	GE	Georgia	MX	Mexico
AU	Australia	GN	Guinea	NE	Niger
BB	Barbados	GR	Greece	NL	Netherlands
BE	Belgium	HU	Hungary	NO	Norway
BF	Burkina Faso	IE	Ireland	NZ	New Zealand
BG	Bulgaria	IT	Italy	PL	Poland
BJ	Benin	JP	Japan	PT	Portugal
BR	Brazil	KE	Kenya	RO	Romania
BY	Belarus	KG	Kyrgyzstan	RU	Russian Federation
CA	Canada	KP	Democratic People's Republic of Korea	SD	Sudan
CF	Central African Republic	KR	Republic of Korea	SE	Sweden
CG	Congo	KZ	Kazakhstan	SG	Singapore
CH	Switzerland	LI	Liechtenstein	SI	Slovenia
CI	Côte d'Ivoire	LK	Sri Lanka	SK	Slovakia
CM	Cameroon	LR	Liberia	SN	Senegal
CN	China	LT	Lithuania	SZ	Swaziland
CS	Czechoslovakia	LU	Luxembourg	TD	Chad
CZ	Czech Republic	LV	Latvia	TG	Togo
DE	Germany	MC	Monaco	TJ	Tajikistan
DK	Denmark	MD	Republic of Moldova	TT	Trinidad and Tobago
EE	Estonia	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	UG	Uganda
FI	Finland	MN	Mongolia	US	United States of America
FR	France	MR	Mauritania	UZ	Uzbekistan
GA	Gabon			VN	Viet Nam

Novel aroylaniline derivatives, compositions and uses thereof in the treatment of viral infections

#### TECHNICAL FIELD OF THE INVENTION

5       The present invention is directed, in general, to aroylaniline derivatives which exhibit antiviral activity, methods for synthesizing these aroylaniline derivatives, pharmaceutical formulations containing these compounds, and methods for treating viral infections, particularly those  
10       caused by a human immunodeficiency virus, such as HIV.

      More particularly, the present invention is related to substantially pure aroylaniline derivatives and related compounds wherein the central double bond is replaced with, for example, a methylene-, ethylene-, hetero atom-, amido-,  
15       ureido- or thioureido-linkage, as well as pharmacological uses and compositions thereof.

#### BACKGROUND OF THE INVENTION

      Only a decade ago, acquired immune deficiency syndrome  
20       (AIDS) was virtually unknown. This puzzling affliction was seen only in a small number of homosexual men. However, today it is difficult to find anyone who has not heard of AIDS, the disease that can debilitate and then kill its victim.

25       AIDS is a fatal disease, reported cases of which have increased dramatically within the past several years. Estimates of reported cases also continue to rise dramatically. Consequently, there is a great need to develop drugs and vaccines to combat AIDS.

30       The virus which is the causative agent of AIDS was first identified in 1983. Since 1983 the AIDS-causing virus has been known by several names and acronyms. Most recently, it has been referred to as human immunodeficiency virus (HIV). It has also been known as lymphadenopathy-  
35       associated virus (LAV) and AIDS-related virus (ARV). HIV is the third known T-lymphocyte virus (HTLV-III), and it has the capacity to replicate within cells of the immune system, causing profound cell destruction. The AIDS virus

is a retrovirus, which is a virus that uses reverse transcriptase during replication. Two distinct families of HIV have been described to date, namely HIV-1 and HIV-2. The acronym HIV will be used herein to refer to HIV viruses generically. Thus far, there is no cure for AIDS.

Specifically, HIV is known to exert a profound cytopathic effect on the CD4+ helper/inducer T-cells, thereby severely compromising the immune system. HIV infection also results in neurological deterioration and, ultimately, in the death of the infected individual.

The field of viral chemotherapeutics has developed in response to the need for agents effective against retroviruses, in particular HIV. There are many ways in which an agent can exhibit anti-retroviral activity. For example, HIV requires at least four viral proteins for replication: reverse transcriptase (RT), protease (PR), transactivator protein (TAT), and regulator of virion-protein expression (REV). Accordingly, viral replication could theoretically be inhibited through inhibition of any one or all of the proteins involved in viral replication. Anti-retroviral agents, such as AZT and ddC, are known to inhibit RT. There also exist anti-retroviral agents that inhibit TAT.

Nucleoside derivatives, such as AZT, are the only clinically active agents that are currently available for antiviral therapy. Although very useful, the utility of AZT and related compounds is limited by toxicity and insufficient therapeutic indices for fully adequate therapy. The development of AZT-resistant strains of HIV also limits the utility of AZT in the treatment of AIDS.

Synthetic peptides also are being developed for potential use as inhibitors of the retroviral PR in the treatment of AIDS. Although these inhibitors are effective in preventing the retroviral PR from functioning, the inhibitors suffer from some distinct disadvantages. First, since the active site of the PR is hindered, i.e., has reduced accessibility as compared to the remainder of the

PR, the ability of the inhibitors to access and bind in the active site of the PR is impaired. Secondly, the peptide inhibitors that bind to the active site of the PR are generally poorly soluble in water, causing distinct problems in drug delivery.

Dyes, particularly azo dyes, have also been discovered to exhibit antiviral activity. For example, compounds which exhibit potential antiherpes activity include the azo dye Trypan Blue (Alarcon et al., "Screening for New Compounds with Antiherpes Activity," Antiviral Res., 4 231-243 (1984); and Thorne et al., "Inactivation of Measles and Herpes Simplex Viruses by Trypan Blue," J. Gen. Virol., 64, 1365-1368 (1983)), as well as Indigocarmine and Paraorange (Westin et al., "Aromatic Sulfonic Acids as Inhibitors: Structure-Activity Study Using Rhino, Adeno 3, Herpes Simplex, and Influenza Viruses," J. Med. Chem., 14, (7), 596-600 (1971)). The azo dye Congo Red and derivatives thereof have also been investigated for potential anti-AIDS activity (Mohan et al., "Potential Anti-AIDS Agents. Synthesis and Anti-viral Activity of Naphthalenesulfonic Acid Derivatives Against HIV-1 and HIV-2," J. Med. Chem., 34, 212-217 (1991)), and the azo dye Evans Blue (Schols et al., "Flow Cytometric Method to Demonstrate Whether Anti-HIV-1 Agent Inhibit Virion Binding to T4+ Cells," J. of Acquired Immune Deficiency Syndromes, 2, 10-15 (1989)). Further, a number of azo dyes were demonstrated to exhibit protective properties in mice infected by equine encephalomyelitis virus (Hurst et al., Brit. J. Pharmacol., 7, 455 (1952)). Eric De Clercq in Anticancer Research, 7, 1023 (1987), and in the Int. J. Cancer, 37, 451 (1986), summarized the perspectives for the chemotherapy of AIDS. These reviews mention two polyanionic dyes with proven anti-HIV activity, i.e., Evans Blue, an azo dye, which is available in 85% purity (Aldrich 1994) and aurintricarboxylic acid which has recently been shown to be a polymeric mixture, and not a monomeric substance. Other anti-virally active azo dyes are

disclosed in U.S. application Serial No. 08/167,296. Thus, dye compounds have been shown to exhibit antiviral activity.

While the above azo dyes have demonstrated anti-HIV activity, they tend to dye the tissues of man or domestic animals. These staining properties have obvious disadvantages. However, the compounds of the present invention were developed based on the discovery that the virucidal activity of such compounds was due to the binding on the uninfected cells, thus preventing attachment/fusion of the virus. Therefore, compounds with similar affinities but lacking the chromophoric azo linkages may express similar biological activity without the disadvantages associated with staining tissue. For example, suramin, an effective agent for the treatment of early cases of human African trypanosomiasis (Burger's Medicinal Chemistry, fourth edition, part II, 444) was developed by replacement of an azo linkage with an amido linkage. Suramin, hexasodium sym-bis(m-aminobenzoyl-m-amino-p-methylbenzoyl-1-naphthyamino-4,6,8-trisulfonate)carbamide, has no azo linkages. The color of suramin is off white (The Merck Index, eleventh edition, 1423). In spite of suramin's disappointing performance in AIDS patients (L.D. Kaplan et al., "Lack of response to suramin in patients with AIDS and AIDS-related complex", Amer. J. Med., 82, 615-20 (1987)), suramin has been investigated in detail (DeClercq, E., "Suramin: A potent inhibitor of the reverse transcriptase of RNA tumor viruses", Cancer Lett., 8:9-22 (1979); Broder, S., Yarchoan R., Collins, J.M. et al., "Effects of suramin on HTLV-III/LAV infection presenting as Kaposi's sarcoma or AIDS-related complex: Clinical pharmacology and suppression of viral replication *in vivo*", Lancet, 2:627-30 (1985); Balzarini, J. Mitsuya H., DeClercq, E., Broder, S., "Comparative inhibitory effects of suramin and other selected compounds on the infectivity and replication of human T-cell lymphotropic virus (HTLV-III)/lymphadenopathy-associated virus (LAV)", Int. J. Cancer, 37:451-7 (1986);

DeClercq, Suramin in the Treatment of AIDS "Mechanism of Action", Antiviral Res., 7, 1-10 (1987); and Nickel, P. et al., "Potentielle Filarizide, Arzneimittel Forschung", 36 (II) #8, 1153-57 (1986). Mohan et al. have extensively  
5 investigated compounds related to suramin. These compounds share the terminal naphthalenesulfonic acid moieties but the substitution pattern of the naphthalenes is different. (Mohan, P. et al., "Naphthalenesulfonic acid derivatives. Potential Anti-AIDS agents", J. Med. Chem., 34 (#1), 212-217  
10 (1991); 35(#26), 4846-4853 (1992), 36(#14), 1996-2003 (1993)).

In addition to the tissue staining properties of azo dyes, such compounds may form metal chelates and are thus difficult to formulate.

15 In view of the above it is desirable to obtain compounds which exhibit biological activity such as virucidal activity with reduced or no staining properties and which are soluble and easier to formulate.

In view of the above, there is a need for compounds  
20 which exhibit biological activity such as virucidal activity with reduced or no staining properties which may be used alone or in combination with AZT and/or other agents in the treatment of viral infections, particularly AIDS. The present invention seeks to provide such  
25 compounds as well as related compositions and methods of using such compositions. These and other objects of the present invention, as well as additional inventive features, will be apparent from the description of the invention provided herein.

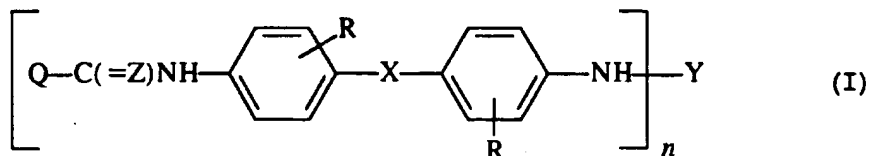
30

#### BRIEF SUMMARY OF THE INVENTION

In accordance with the present invention, it has now been discovered that certain compounds, specifically, aroylaniline derivatives as defined in Formula I below, are  
35 capable of preventing the replication of a virus in a cell, such as a human T-cell, infected with such a virus, without staining the tissue. Also, infections of mammals,

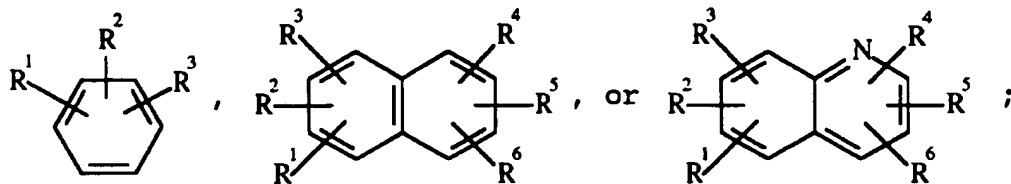
particularly humans, by a virus, particularly a retrovirus such as a human immunodeficiency virus, may be effectively treated with the compounds of Formula I.

The aroylaniline derivatives and related compounds of the present invention have the following formula:

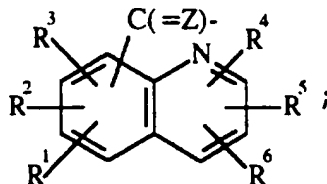
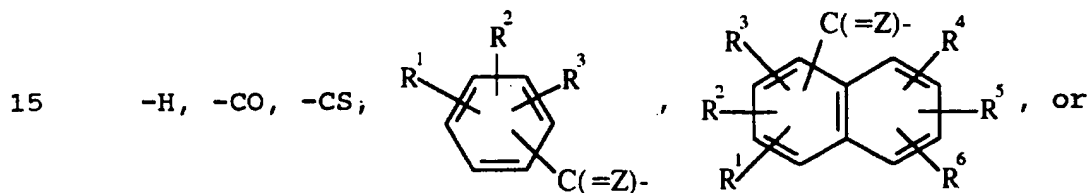


wherein R is -H, -SO<sub>3</sub>·E, -CO<sub>2</sub>·E, -PO(O)<sub>2</sub>·2E, -NO<sub>2</sub>, or a halogen;

Q is



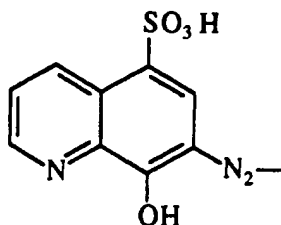
Y is





7

$R^1$ ,  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^5$  and  $R^6$  are each independently -H, -OH, a halogen,  $-NH_2$ ,  $-NHCH_3$ ,  $-N(CH_3)_2$ ,  $-CHO$ ,  $CH_3O-$ ,  $-SO_3 \cdot E$ ,  $-NO_2$ ,  $-CO_2 \cdot E$ ,  $-PO(O)_2 \cdot 2E$ ,  $-CH_2OH$ ,  $-CH_2SCH_2COR^7$ ,  $-CH_2S-$   
 5  $C_2H_4COR^7$ ,  $-CH_2S-CH(CO_2H)CH_2COR^7$ ,  $-CH_2S-CH(CO_2H)CH_2CO_2R^7$ ,  
 $-CH_2SCH_2CO_2R^7$ ,  $-CH_2SC_2H_4CO_2R^7$  or



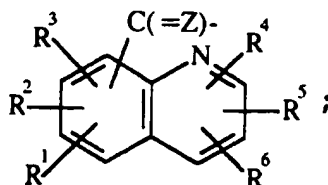
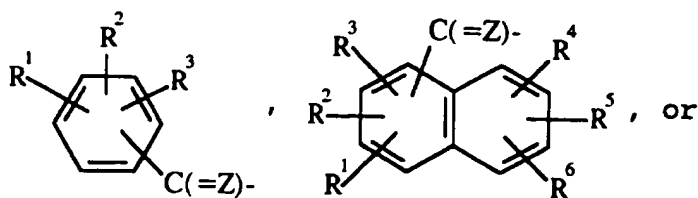
wherein E represents H, lithium, sodium, potassium, ammonium, magnesium,  $(HOC_2H_4)_3NH$ ,  $(CH_3)_3NH$ ,  $CH_3NH(C_2H_4OH)_2$ ,  $(CH_3)_4N$ , or HN-methylglucamine and wherein  $R^7$  is -H, methyl,  
 10 ethyl, benzyl,  $-NH_2$ ,  $-NHCH_3$ , or  $-NHC_2H_5$ ;

X is  $-CHCH-$ ,  $-CH_2CH_2-$ ,  $-CONH-$ ,  $-NHCONH-$ ,  $-NHCSNH-$ ,  $-S-$ ,  $-SO-$ ,  $-SO_2-$ ,  $-NH-$ ,  $-CH_2-$  or cyclopropyl;

Z is oxygen or sulfur; and

n is 1 or 2, but when n is 1, Y is H,

15



and when n is 2, Y is  $-CO$  or  $-CS$ .

The present invention also provides for a method inhibiting replication of a virus, particularly a retrovirus, and more particularly a human immunodeficiency virus, specifically HIV-1 or HIV-2 both *in vitro* and *in vivo*, by contacting cells with the aroylaniline derivatives of the present invention.

The present invention further provides a method of treating a mammal, particularly a human, infected with a virus, particularly a retrovirus such as a human immunodeficiency virus, comprising administering a retrovirally effective amount of an aroylaniline derivative of Formula I. Additionally, the present invention provides a method of preventing a mammal, particularly a human, from being infected with a virus, particularly a retrovirus such as a human immunodeficiency virus, comprising administering a prophylactically effective amount of an aroylaniline derivative of Formula I.

The present invention also provides a method for utilizing the aroylaniline derivatives of Formula I for screening and other diagnostic testing.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 depicts *in vitro* testing results for anti-HIV activity for the present inventive compound of Example 1.

Figure 2 depicts *in vitro* testing results for anti-HIV activity for the present inventive compound of Example 3.

Figure 3 depicts *in vitro* testing results for anti-HIV activity for the present inventive compound of Example 13.

Figures 4 depicts *in vitro* testing results for anti-HIV activity for the present inventive compound of Example 14.

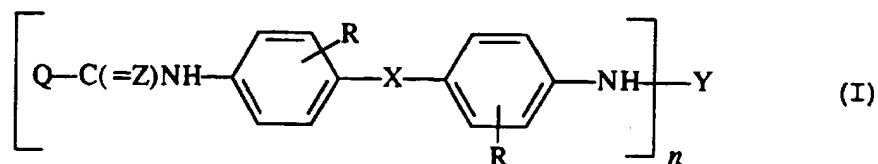
Figure 5 depicts *in vitro* testing results for anti-HIV activity for the present inventive compound of Example 29.

Figure 6 depicts *in vitro* testing results for anti-HIV activity for the present inventive compound of Example 32.

## DETAILED DESCRIPTION OF THE INVENTION

The aroylaniline derivatives and related compounds of the present invention have the following formula:

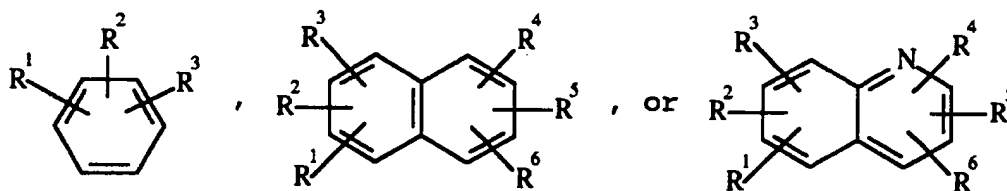
5



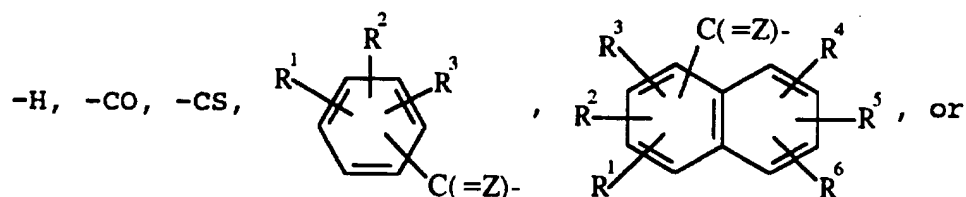
wherein R is -H, -SO<sub>3</sub>'E, -CO<sub>2</sub>'E, -PO(O)<sub>2</sub>'2E, -NO<sub>2</sub>, or a halogen;

Q is

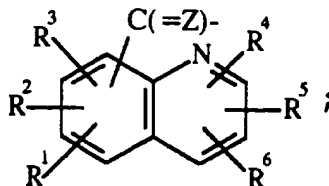
10



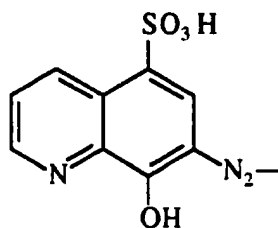
Y is



15



- $R^1, R^2, R^3, R^4, R^5$  and  $R^6$  are each independently -H, -OH, a halogen,  $-NH_2$ ,  $-NHCH_3$ ,  $-N(CH_3)_2$ ,  $-CHO$ ,  $CH_3O-$ ,  $-SO_3 \cdot E$ ,  $-NO_2$ ,  $-CO_2 \cdot E$ ,  $-PO(O)_2 \cdot 2E$ ,  $-CH_2OH$ ,  $-CH_2SCH_2COR^7$ ,  $-CH_2S-$   
 5  $C_2H_4COR^7$ ,  $-CH_2S-CH(CO_2H)CH_2COR^7$ ,  $-CH_2S-CH(CO_2H)CH_2CO_2R^7$ ,  
 $-CH_2SCH_2CO_2R^7$ ,  $-CH_2SC_2H_4CO_2R^7$  or

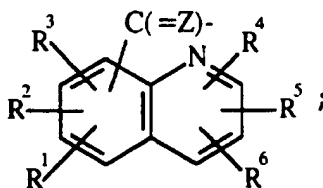
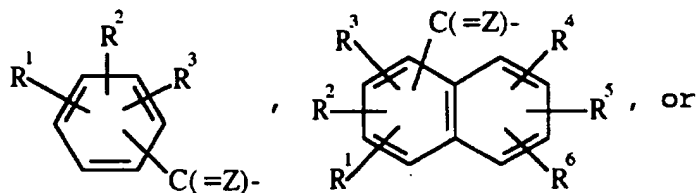


- wherein E represents H, lithium, sodium, potassium, ammonium, magnesium,  $(HOC_2H_4)_3NH$ ,  $(CH_3)_3NH$ ,  $CH_3NH(C_2H_4OH)_2$ ,  
 10 or HN-methylglucamine and wherein  $R^7$  is -H, methyl, ethyl, benzyl,  $-NH_2$ ,  $-NHCH_3$ , or  $-NHC_2H_5$ ;

X is  $-CHCH-$ ,  $-CH_2CH_2-$ ,  $-CONH-$ ,  $-NHCONH-$ ,  $-NHCSNH-$ ,  $-S-$ ,  $-SO-$ ,  $-SO_2-$ ,  $-NH-$ ,  $-CH_2-$  or cyclopropyl;

Z is oxygen or sulfur; and

- 15 n is 1 or 2, but when n is 1, Y is H,



and when n is 2, Y is  $-CO$  or  $-CS$ .

Preferably, the aroylaniline derivatives have the structural formulas indicated by the compounds identified in Table 1.

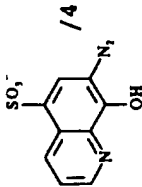
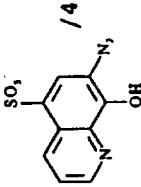
TABLE 1

COMPOUND	X/+	Q	Z	POSITION OF AMIDE	R/+	R <sup>1</sup> /+	R <sup>2</sup> /+	R <sup>3</sup> /+	R <sup>4</sup> /+	R <sup>5</sup> /+	R <sup>6</sup> /+	Y	n	SALT	
5	1	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	H	H	H	U	1	Na
	2	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	H	H	H	H	1	Na
	3	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	H	H	H	CO	2	Na
	4	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	H	H	H	H	H	U	1	Na
10	5	SO <sub>2</sub> /1	U	O	7	CO <sub>2</sub> <sup>-</sup> /3	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	H	H	H	U	1	K
	6	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CHO/5	H	H	H	H	U	1	Na
	7	CH <sub>2</sub> /1	U	O	7	CO <sub>2</sub> <sup>-</sup> /3	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	H	H	H	U	1	NH <sub>4</sub>
	8	SO <sub>2</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	Cl/2	H	H	U	1	Li
15	9	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	Cl/4	H	H	U	1	(CH <sub>3</sub> ) <sub>3</sub> NH
	10	C <sub>2</sub> H <sub>4</sub> /2	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> OH/5	H	H	H	H	U	1	NH <sub>4</sub>
	11	NHCOHN/1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	SO <sub>3</sub> <sup>-</sup> /5	H	OCH <sub>3</sub> /4	H	H	U	1	K
	12	1,2-cyclo- propyl/1	U	O	7	CO <sub>2</sub> <sup>-</sup> /2	OH/8	Cl/5	H	H	H	H	U	1	(CH <sub>3</sub> ) <sub>4</sub> N
13	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SC <sub>2</sub> H <sub>4</sub> CO <sub>2</sub> <sup>-</sup> /5	H	H	H	H	U	1	Na	

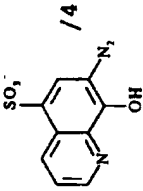
COMPOUND	X/+	0	2	POSITION OF AMIDE	R/+	R <sup>1</sup> /+	R <sup>2</sup> /+	R <sup>3</sup> /+	R <sup>4</sup> /+	R <sup>5</sup> /+	R <sup>6</sup> /+	Y	n	SALT
14	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SCH <sub>2</sub> CO <sub>2</sub> <sup>-</sup> /5	H	H	H	H	U	1	Na
15	C <sub>2</sub> H <sub>4</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> S-CH(CO <sub>2</sub> <sup>-</sup> ) CH <sub>2</sub> CO <sub>2</sub> <sup>-</sup> /5	H	H	H	H	U	1	(HOC <sub>2</sub> H <sub>4</sub> ) <sub>3</sub> N H
16	C <sub>2</sub> H <sub>5</sub> /1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> S-CH <sub>2</sub> CO <sub>2</sub> CH <sub>3</sub> /5	H	H	H	H	U	1	NH <sub>4</sub>
17	1,2-cyclo- propyl/1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> S-CH <sub>2</sub> CONH <sub>2</sub> /5	H	H	H	H	U	1	Na
18	S/1	U	O	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SCH <sub>2</sub> CONHCH <sub>3</sub> /5	H	H	H	H	U	1	Na
19	C <sub>2</sub> H <sub>5</sub> /1	U	S	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	H	H	H	H	H	U	1	Na
20	C <sub>2</sub> H <sub>5</sub> /1	U	S	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SCH <sub>2</sub> CO <sub>2</sub> <sup>-</sup> /5	H	H	H	H	U	1	NH <sub>4</sub>
21	C <sub>2</sub> H <sub>4</sub> /1	U	S	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SC <sub>2</sub> H <sub>4</sub> CO <sub>2</sub> <sup>-</sup> /5	H	H	H	H	U	1	NH <sub>4</sub>
22	SO <sub>2</sub> /1	U	S	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SCH <sub>2</sub> CONH <sub>2</sub> /5	H	H	H	H	U	1	Li
23	S/1	U	S	7	SO <sub>3</sub> <sup>-</sup> /2	OH/8	CH <sub>2</sub> SCH <sub>2</sub> CO <sub>2</sub> CH <sub>3</sub> /5	H	H	H	H	U	1	Li
24	C <sub>2</sub> H <sub>5</sub> /1	N	O	2	SO <sub>3</sub> <sup>-</sup> /2	OH/1	CH <sub>2</sub> SC <sub>2</sub> H <sub>4</sub> CO <sub>2</sub> <sup>-</sup> /4	H	H	H	H	N	1	Na

5

10

COMPOUND	X/+	Q	Z	POSITION OF AMIDE	R/+	R <sup>1</sup> /+	R <sup>2</sup> /+	R <sup>3</sup> /+	R <sup>4</sup> /+	R <sup>5</sup> /+	R <sup>6</sup> /+	Y	n	SALT	
25	C <sub>2</sub> H <sub>4</sub> /1	N	O	2	SO <sub>3</sub> <sup>-</sup> /2	OH/1	CH <sub>2</sub> SCH <sub>2</sub> CO <sub>2</sub> <sup>-</sup> /4	H	H	H	H	N	1	NH <sub>4</sub>	
26	C <sub>2</sub> H <sub>2</sub> /1	N	O	2	SO <sub>3</sub> <sup>-</sup> /2	OH/1	CH <sub>2</sub> SCH <sub>2</sub> CO <sub>2</sub> CH <sub>3</sub> /4	H	H	H	H	N	1	Na	
27	C <sub>2</sub> H <sub>4</sub> /1	N	O	2	SO <sub>3</sub> <sup>-</sup> /2	OH/1	CH <sub>2</sub> SCH <sub>2</sub> CONH <sub>2</sub> /4	H	H	H	H	H	1	(CH <sub>3</sub> ) <sub>4</sub> N	
28	C <sub>2</sub> H <sub>2</sub> /1	N	O	2	SO <sub>3</sub> <sup>-</sup> /2	OH/1	CH <sub>2</sub> SC <sub>2</sub> H <sub>4</sub> <sup>-</sup> CONHCH <sub>3</sub> /4	H	H	H	H	H	1	Na	
29	C <sub>2</sub> H <sub>2</sub> /1	P	O	1	SO <sub>3</sub> <sup>-</sup> /2	CO <sub>2</sub> <sup>-</sup> /2	OH/3	OH/6	---	---	---	P	1	Na	
30	C <sub>2</sub> H <sub>2</sub> /1	P	O	1	CO <sub>2</sub> <sup>-</sup> /2	CO <sub>2</sub> <sup>-</sup> /2	OH/3	OH/6	---	---	---	P	1	Na	
31	C <sub>2</sub> H <sub>2</sub> /1	P	O	1	CO <sub>2</sub> <sup>-</sup> /3	CO <sub>2</sub> <sup>-</sup> /2	OH/3	OH/6	---	---	---	P	1	K	
10	32	C <sub>2</sub> H <sub>2</sub> /1	P	O	1	SO <sub>3</sub> <sup>-</sup> /2	SO <sub>3</sub> <sup>-</sup> /2			H	---	---	P	1	Na
33	C <sub>2</sub> H <sub>2</sub> /1	P	O	1	CO <sub>2</sub> <sup>-</sup> /2	SO <sub>3</sub> <sup>-</sup> /2			H	---	---	P	1	Na	

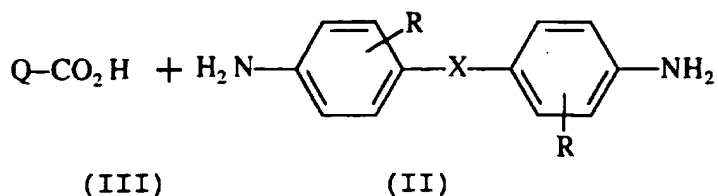


COMPOUND	X/+	Q	Z	POSITION OF AMIDE	R/+	R <sup>1</sup> /+	R <sup>2</sup> /+	R <sup>3</sup> /+	R <sup>4</sup> /+	R <sup>5</sup> /+	R <sup>6</sup> /+	Y	n	SALT
34	C <sub>6</sub> H <sub>4</sub> /1	P	O	1	SO <sub>3</sub> <sup>-</sup> /2	SO <sub>3</sub> <sup>-</sup> /2		H	---	---	---	P	1	Na

5 Notes: U = QUINOLINE; N = NAPHTHLENE; P = PHENYL; + = POSITION

Th aroylaniline derivatives of Formula I, particularly compounds 1-12 in Table 1, are prepared by reacting the requisite aromatic acid (III) with the aniline derivative (II).

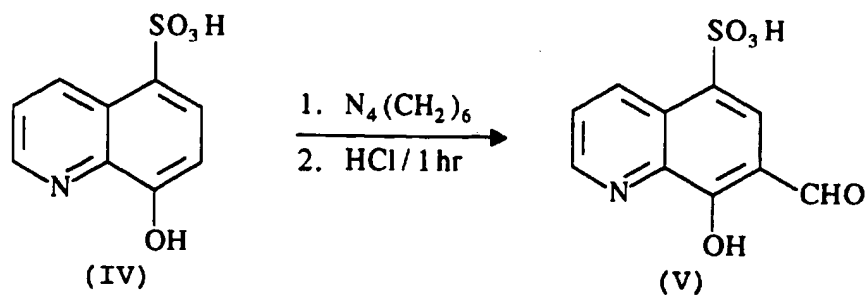
5



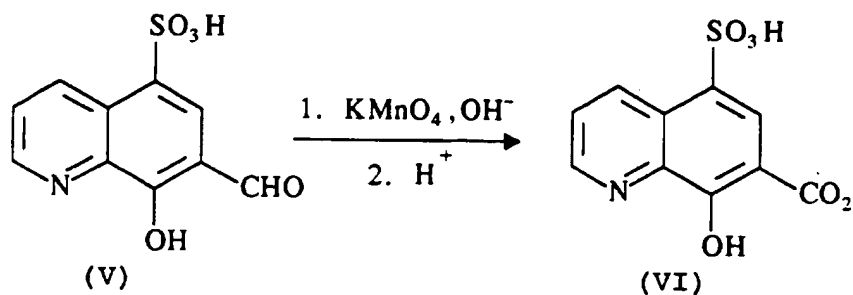
Preferably, the aromatic acid (III) is first activated  
10 by converting the acid into a symmetric anhydride, mixed anhydride, active ester, or acid chloride, by using N,N-diethylclohexylcarbodiimide (DCC) and N-hydroxybenzotriazole (HBT) as activating agents. Additional routes toward amides are well documented in Barton's Comprehensive  
15 Organic Chemistry, vol. II, 958, Pergamon 1979, and in T.W. Greene's Protective Groups in Organic Chemistry, J. Wiley & Sons, N.Y. 1991. When using DCC and HBT as coupling agents, DMF is the preferred solvent. The reaction can be conducted at elevated temperature for shorter periods, but  
20 room temperature is preferable. Reaction times may vary from one to eight days. To assure optimal yields of the desired aroylaniline derivatives and a minimum of by-products, the reaction is preferably conducted for a maximum of eight days.

25 The key intermediate for quinolinoyl-substituted anilines is prepared starting with the commercially available 8-hydroxyquinoline-5-sulfonic acid (IV), which is formylated for example using the Duff reaction to furnish 8-hydroxyquinoline-7-formyl-5-sulfonic acid (V). Potassium  
30 permanganate oxidation of the product under basic conditions yields 8-hydroxyquinoline-5-sulfo-7-carboxylic acid (VI) which is then coupled with an aniline derivative (II) using HBT and DCC as activating agents to furnish the quinoline compounds of Formula I.

17

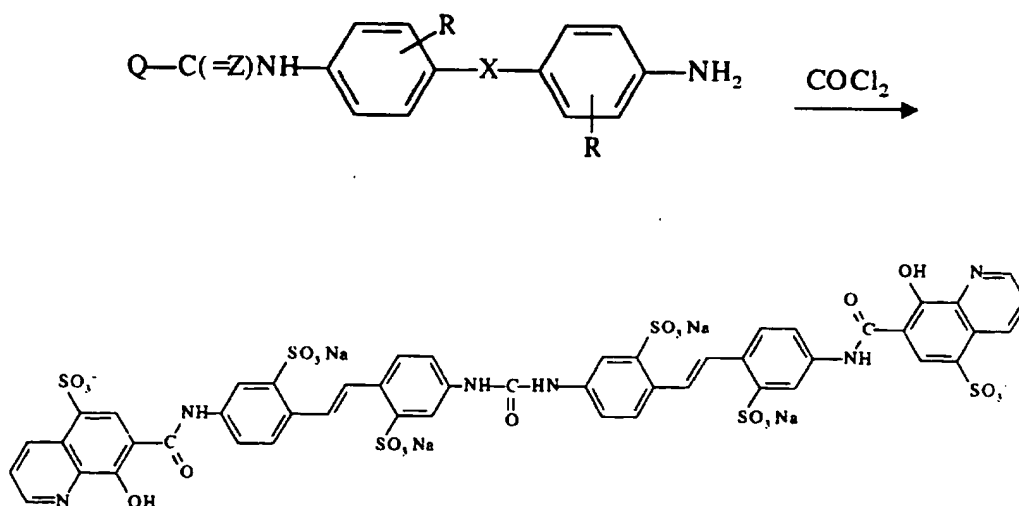


5



10        Compounds of Formula I, wherein n is 2 and Y is -CO, such as Compound 3 in Table 1, are synthesized by carbonylation of compounds I, where n is 1 and Y is H, using either phosgene, triphosgene or carbonyldiimidazole as a reagent, as shown below for the synthesis of compound

15    3.



18

Thio-Mannich products, i.e., Formula I, (where n is 1, R<sup>2</sup> is -CH<sub>2</sub>SCH<sub>2</sub>COR<sup>7</sup>, -CH<sub>2</sub>S-C<sub>2</sub>H<sub>4</sub>COR<sup>7</sup>, -CH<sub>2</sub>S-CH(CO<sub>2</sub>H)CH<sub>2</sub>COR<sup>7</sup>, CH<sub>2</sub>S-CH(CO<sub>2</sub>H)CH<sub>2</sub>CO<sub>2</sub>R<sup>7</sup>, and Y is phenyl, quinoline or naphthalene ring structure as defined above) are obtained  
 5 by reacting compounds of Formula I, wherein at least one R<sup>1</sup>-R<sup>6</sup> substituent is a hydroxy group with a hydrogen at its ortho or para position (such as compounds 13-18 in Table 1), with formaldehyde and the appropriate mercaptoalkanoic acid derivative (VII). The reaction temperature may vary  
 10 from room temperature to 100°C. The reaction time may vary from one to ten hours. The preferred time is about three hours at 100°C.

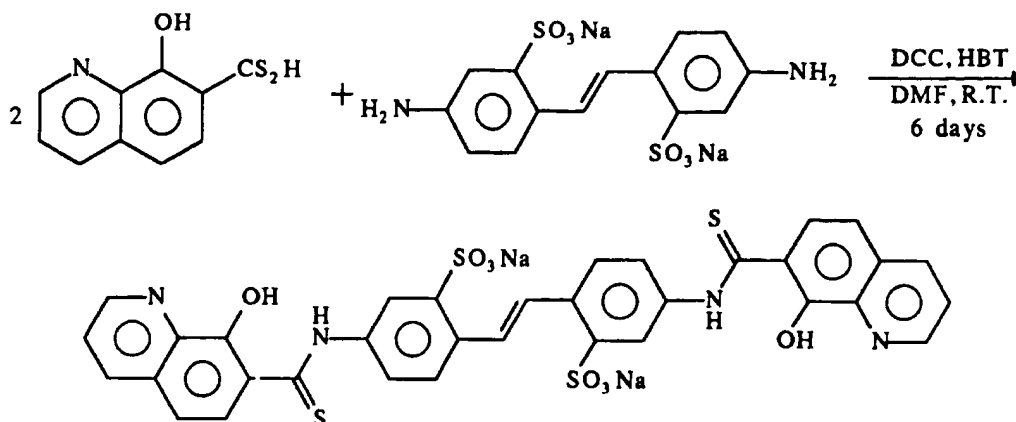


where acid derivative (VII) is HSCH<sub>2</sub>COR<sup>7</sup>, HSC<sub>2</sub>H<sub>4</sub>COR<sup>7</sup>, HSCH(CO<sub>2</sub>H)CH<sub>2</sub>COR<sup>7</sup>, HSCH<sub>2</sub>CO<sub>2</sub>R<sup>7</sup>, HSC<sub>2</sub>H<sub>4</sub>CO<sub>2</sub>R<sup>7</sup>,  
 20 HSCH(CO<sub>2</sub>H)CH<sub>2</sub>CO<sub>2</sub>R<sup>7</sup>, and R<sup>2</sup> is -CH<sub>2</sub>SCH<sub>2</sub>COR<sup>7</sup>, -CH<sub>2</sub>S-C<sub>2</sub>H<sub>4</sub>COR<sup>7</sup>, -CH<sub>2</sub>S-CH(CO<sub>2</sub>H)CH<sub>2</sub>COR<sup>7</sup>, CH<sub>2</sub>S-CH(CO<sub>2</sub>H)CH<sub>2</sub>CO<sub>2</sub>R<sup>7</sup>, -CH<sub>2</sub>SCH<sub>2</sub>CO<sub>2</sub>R<sup>7</sup>, -CH<sub>2</sub>SC<sub>2</sub>H<sub>4</sub>CO<sub>2</sub>R<sup>7</sup>.

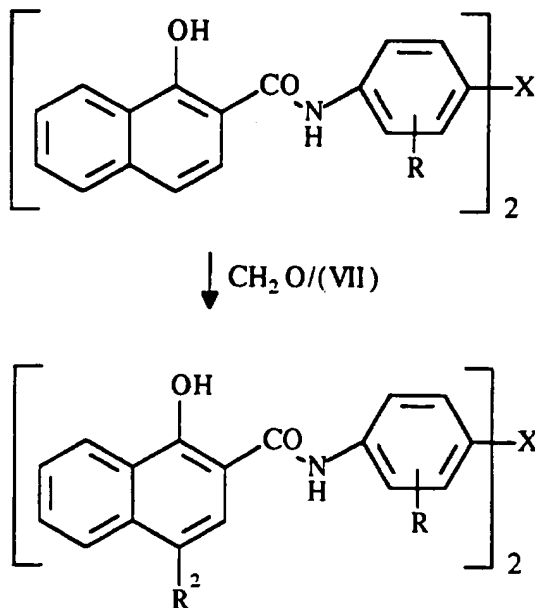
Quinoline substituted compounds (I), which have a thioamide (CSNH) linkage, are prepared by reacting 8-  
 25 hydroxyquinoline with o-ethylxanthic acid (potassium salt) at elevated temperatures in an autoclave to obtain 8-hydroxyquinoline-7-thioacid. The preferred temperature range is 130-180°C. Coupling of this compound to aniline derivatives (II), in the presence of DCC/HBT yields  
 30 thioamide linked quinolines, such as Compounds 19-23 in Table 1. The last step of the sequence is illustrated by the preparation of Compound 19.

Thioamides may also be prepared by reacting amides with the dimer of p-m thoxyphenylthiono phosphine sulfide,  
 35 Lawesson's Reagent.

19



Compounds (I) which have hydroxynaphthalene groups as terminal substituents, such as Compounds 24-28 in Table 1, can be reacted with mercaptoalkanoic acids, esters or amides and formalin to give the corresponding thio-Mannich products with the carboxy-, alkoxy-carbonyl- or amidoalkylthiomethyl group at the ortho(para) position (relative to the hydroxy group). The reaction time may vary from one to twelve hours at temperatures ranging from 50 to 100°C.

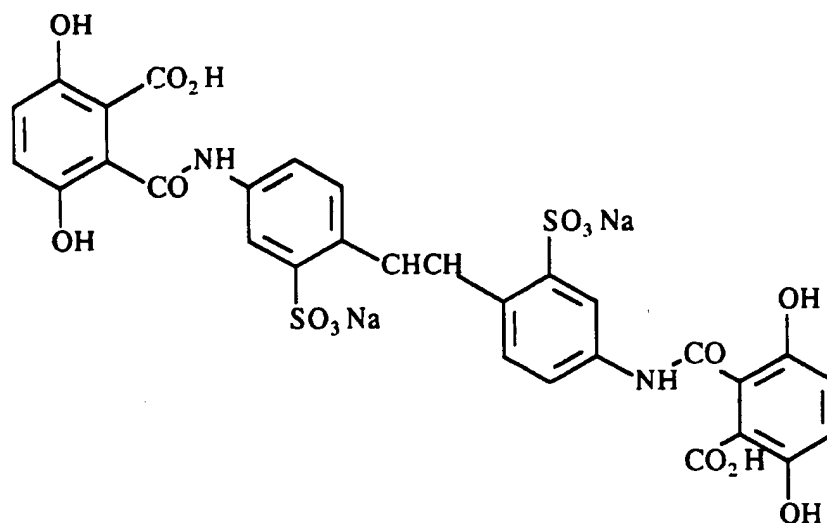


where VII is  $\text{HSCH}_2\text{COR}^7$ ,  $\text{HSC}_2\text{H}_4\text{COR}^7$ ,  $\text{HSCH}(\text{CO}_2\text{H})\text{CH}_2\text{COR}^7$ ,  $\text{HSCH}_2\text{CO}_2\text{R}^7$ ,  $\text{HSC}_2\text{H}_4\text{CO}_2\text{R}^7$ ,  $\text{HSCH}(\text{CO}_2\text{H})\text{CH}_2\text{CO}_2\text{R}^7$ , and  $\text{R}^2$  is

20

$-\text{CH}_2\text{SCH}_2\text{COR}^7$ ,  $-\text{CH}_2\text{S}-\text{C}_2\text{H}_4\text{COR}^7$ ,  $-\text{CH}_2\text{S}-\text{CH}(\text{CO}_2\text{H})\text{CH}_2\text{COR}^7$ ,  $\text{CH}_2\text{S}-\text{CH}(\text{CO}_2\text{H})\text{CH}_2\text{CO}_2\text{R}^7$ ,  $-\text{CH}_2\text{SCH}_2\text{CO}_2\text{R}^7$ , or  $-\text{CH}_2\text{SC}_2\text{H}_4\text{CO}_2\text{R}^7$ .

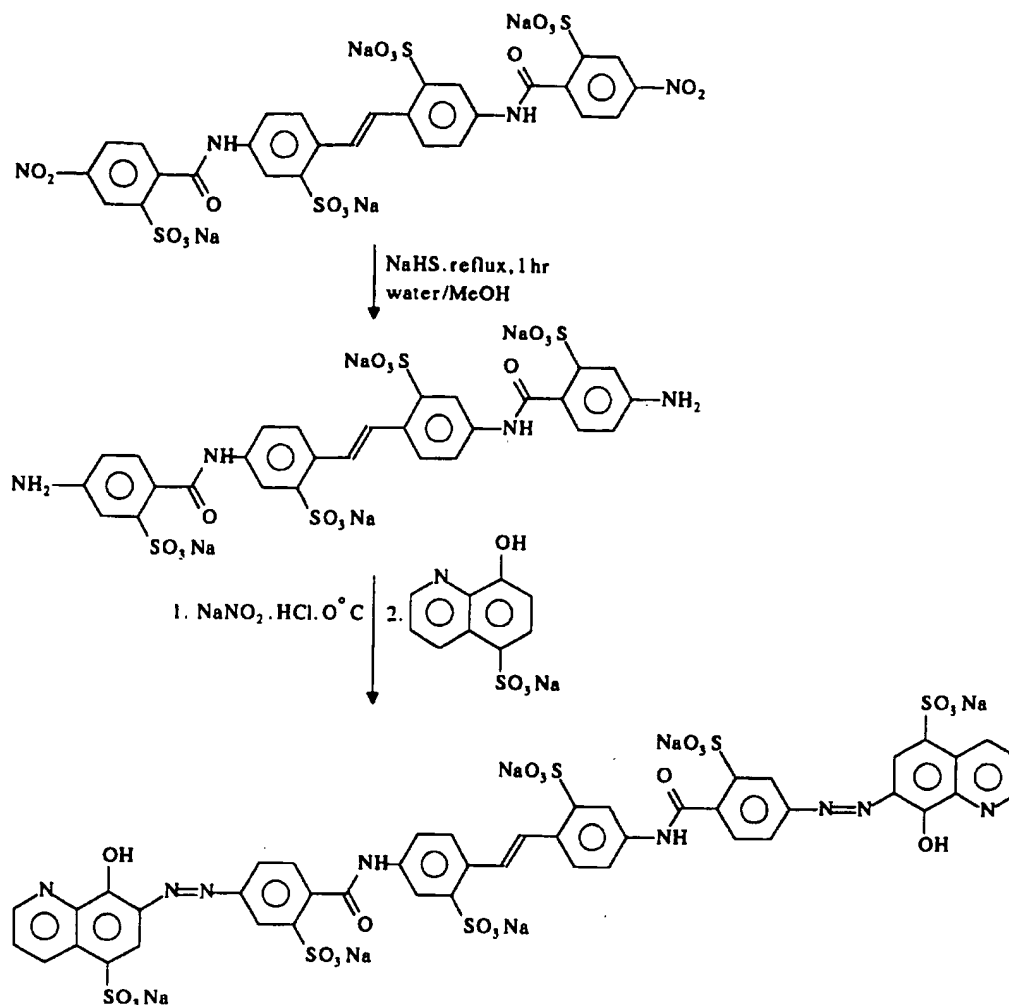
Acylation of 4,4'-diaminostilbene-2,2'-disulfonic acid disodium salt with 2,3-dihydroxyphthalic anhydride  
 5 furnishes the ring-opened derivative, Compound 29 in Table 1, depicted below.



10

The synthesis of Compound 32 in Table 1 is depicted below.

21



5 The aroylaniline derivatives of the present invention act as a chemical barrier to retroviral, particularly HIV, transmission. Specifically, it is believed that the mechanism of action is that the aroylaniline derivatives bind to the CD4+ amino acid region that contains a high concentration of arginine which retroviruses such as HIV

10 typically bind to.

The aroylaniline derivatives of the present invention can be used in the form of their pharmaceutically acceptable salts, can be used alone or in appropriate association, and can also be used in combination with other

15 pharmaceutically active compounds. The active agent can be

present in the pharmaceutical composition in any suitable quantity. The pharmaceutical excipients described herein, for example, vehicles, adjuvants, carriers or diluents, are well known to those who are skilled in the art, and readily  
5 available. The choice of excipient will be determined in part by the particular compound, as well as by the particular method used to administer the composition. Accordingly, there is a wide variety of suitable formulations of the pharmaceutical compositions of the  
10 present invention. The following methods and excipients are merely exemplary and are in no way limiting.

The aroylaniline derivatives of the present invention can be used *in vitro* for diagnostic testing and other assays or screening methods well known to one of ordinary  
15 skill in the art. For example, the compounds may be used for studying viruses, particularly retroviruses.

While the method of the present invention can be practiced *in vitro*, it has particular usefulness in *in vivo* applications. As regards these applications, the present  
20 invention includes the administration to a mammal, particularly a human, of a virally effective amount, particularly a retrovirally effective amount of one or more of the aforementioned aroylaniline derivatives as the active agent effective against a viral infection, such as  
25 AIDS and AIDS-related infections.

One skilled in the art will appreciate that suitable methods of administering a compound of the present invention to a mammal are available, and, although more than one route can be used to administer a particular  
30 compound, a particular route can provide a more immediate and more effective reaction than another route. The compounds may be administered in any suitable manner, including orally, buccally, sublingually, subcutaneously, intramuscularly, intravenously or rectally.

35 The aroylaniline derivatives of the present invention can be employed in pharmaceutical compositions and used for



treating viral infections, particularly retroviral infections. For example, the present inventive aroylaniline derivatives can be used to treat infections by viruses, such as Hepatitis B, HTLV-1, HTLV-II, HTLV-IV, HTLV-V, HIV-1, HIV-2, HIV-3 and HIV-4, herpes simplex I, herpes simplex II, herpes zoster, Epstein-Barr virus, cytomegalo viruses and influenza.

The aroylaniline derivatives employed in the present invention can be made into pharmaceutical compositions by combination with appropriate pharmaceutically acceptable carriers or diluents, and may be formulated into preparations in solid, semi-solid, liquid or gaseous forms such as tablets, capsules, granules, powders, solutions, suppositories, injections, inhalants, and aerosols in the usual ways for their respective route of administration.

In the case of oral preparations, the aroylaniline derivatives can be used alone or in combination with appropriate additives to make tablets, powders, granules or capsules, e.g., with conventional additives such as lactose, mannitol, corn starch or potato starch; with binders such as crystalline cellulose, cellulose derivatives, acacia, corn starch or gelatins; with disintegrators such as corn starch, potato starch or sodium carboxymethylcellulose; with lubricants such as talc or magnesium stearate; and if desired, with diluents, buffering agents, moistening agents, preservatives and flavoring agents.

Furthermore, the aroylaniline derivatives employed in the present invention can be made into suppositories by mixing with a variety of bases such as emulsifying bases or water-soluble bases. The suppository can include vehicles such as cocoa butter, carbowaxes and polyethylene glycols, which melt at body temperature, yet are solid at room temperature.

Formulations suitable for parenteral administration include aqueous and non-aqueous, isotonic, and sterile

injection solutions. The aroylaniline derivatives employed in the present invention can be formulated into preparations for injections by dissolving, suspending or emulsifying them in an aqueous or non-aqueous solvent, such as vegetable oil, synthetic aliphatic acid glycerides, esters of higher aliphatic acids or propylene glycol, and, if desired, with conventional additives such as solubilizers, isotonic agents, suspending agents, emulsifying agents, stabilizers, thickening agents and preservatives. Solution forms can include buffers, such as acetate and phosphate, toxicity adjusting agents, such as sodium chloride, pH adjusting agents, such as HCl and phosphoric acid, bactericides, solutes that render the solution isotonic with the blood of the intended recipient, and the like.

Extracorporeal injection solutions and suspensions can be prepared from sterile powders, granules and tablets of the kind previously described.

In the case of inhalations or aerosol preparations, the aroylaniline derivatives employed in the invention in the form of a liquid or minute powder can be utilized in an aerosol container with gas or liquid propellants, such as dichlorofluoromethane, propane, nitrogen, and the like, and, if desired, together with conventional adjuvants such as humidifying agents. They can also be formulated as pharmaceuticals for non-pressured preparations such as in a nebulizer or an atomizer.

Unit dosage forms for oral administration such as syrups, elixirs, and suspensions wherein each dosage unit, e.g., teaspoonful or tablespoonful, contains a predetermined amount of the aroylaniline derivatives employed in the present invention can be by a pharmaceutically acceptable carrier, such as Sterile Water for Injection, USP, or normal saline.

The term "unit dosage form" as used herein refers to physically discrete units suitable as unitary dosages for

human and animal subjects, each unit containing a predetermined quantity of the aroylaniline derivatives calculated in an amount sufficient to produce the desired effect in association with a pharmaceutically acceptable, diluent, carrier or vehicle. The specifications for the novel unit dosage forms of the present invention depend on the particular compound employed and the effect to be achieved, and the pharmacodynamics associated with each compound in the host.

The amount of aroylaniline derivative of the present invention to be used varies according to the degree of infection encountered and the stage of the disease. A suitable dosage is that which will result in concentration of the aroylaniline derivative in blood and/or tissue harboring virus which is known to inhibit the virus, e.g., about  $10^{-9}$  M to  $10^{-4}$  M, more preferably  $10^{-8}$  M to  $10^{-6}$  M. The preferred dosage is that amount sufficient to beneficially affect a host and possibly render the host asymptomatic to the particular viral infection. For example, the dosage amount can vary between 0.01-5 mg/kg body weight administered 1-5 times daily. The dose may vary when the compounds are used prophylactically.

Any necessary adjustments in dose can be readily made to meet the severity of the infection and adjusted accordingly by the skilled practitioner.

The following examples further illustrate the present invention but, of course, should not be construed as in any way limiting its scope.

30

#### EXAMPLES

##### Materials and Methods for Synthesizing Aroylaniline Compounds

In each of the following examples, melting points were determined with a Kofler hot stage microscope and are corrected. All temperatures are in degrees centigrade. NMR spectra were determined with a Varian Gemini-300 (300 MHz) spectrometer. Chemical shifts ( $\delta$ ) were reported

relative to the appropriate deuterated standard. Mass spectra were recorded on a VG Analytical 70-SE mass spectrometer equipped with an 11-205 J data system. IR spectra were recorded on a Nicolet 520 FT-IR spectrometer. UV spectra were recorded on a Shimadzu UV-VIS scanning spectrometer. Elemental analysis was performed by Atlantic Microlab, Inc., Norcross, Georgia.

#### EXAMPLE 1

This Example illustrates the synthesis of the aroylaniline derivative, 4,4'-di(8-hydroxy-5-sulfo-7-quinolinecarboxyamido)stilbene-2,2'-disulfonic acid tetrasodium salt, Compound 1 as shown in Table 1.

#### A. Synthesis of 7-formyl-8-hydroxyquinoline-5-sulfonic acid.

8-Hydroxyquinoline-5-sulfonic acid monohydrate (0.015 mol, 3.65 g), hexamethylenetetramine (0.225 mol, 21.0 g) and acetic acid (750 mL) were stirred and heated at 97-100°C for 6 hrs. After cooling, acetic acid was removed under reduced pressure, and the residue was treated with acetone (1 L) to give a precipitate which was filtered off and washed with additional acetone. Water (30 mL) and concentrated hydrochloric acid (15 mL) were then added to the solid. The resulting solution was refluxed for 2 hours. Then, the mixture was concentrated under reduced pressure and left for crystallization. The solid was filtered off, washed with water and dried to give pure 7-formyl-8-hydroxyquinoline-5-sulfonic acid in 45% yield.

m.p. 286-288°C

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ [ppm]:

10.54 (s, 1H, H-e)

9.22 (dd, J=8.7; 1.5 Hz, 1H, H-a)

8.99 (dd, J=4.5; 1.5 Hz, 1H, H-c)

8.17 (s, 1H, H-d)

7.84 (dd, J=8.7; 4.5 Hz, 1H, H-b)

IR (KBr)  $\nu_{\max}$  [ $\text{cm}^{-1}$ ]:

3442, 3109, 2926, 2855, 1657, 1611, 1594, 1552, 1401, 1330, 1282, 1244, 1176, 1126, 1045, 987, 942, 897, 828, 765, 743, 634, 591.

5

B. Oxidation of 7-formyl-8-hydroxyquinoline-5-sulfonic acid

The 7-formyl-8-hydroxyquinoline-5-sulfonic acid (15 mmol, 3.80 g) was dissolved in sodium carbonate solution (34.5 mmol, 3.66 g in 200 mL of  $\text{H}_2\text{O}$ ). The solution was stirred and heated at  $52^\circ\text{C}$  (oil bath). Then to the hot solution potassium permanganate (37.5 mmol, 5.94 g) was added. The temperature was increased to  $68^\circ\text{C}$ . The stirring and heating ( $60$ - $65^\circ\text{C}$ ) was continued for 55 minutes. After that time the reaction mixture was filtered, and the solid was washed with sodium carbonate solution followed by water. The filtrate was concentrated under reduced pressure, acidified with 50% sulfuric acid (pH 3) and left for crystallization. The precipitate was filtered off, washed with a small amount of water and dried to give 1.96 g of 7-carboxy-8-hydroxyquinoline-5-sulfonic acid (48.5%).

20

m.p.  $289$ - $291^\circ\text{C}$

$^1\text{H}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$  [ppm]:

9.39 (dd,  $J=8.7$ ; 1.5 Hz, 1H, H-a)

25 8.89 (dd,  $J=4.5$ ; 1.5 Hz, 1H, H-c)

8.33 (s, 1H, H-d)

7.88 (dd,  $J=8.7$ ; 4.5 Hz, 1H, H-b)

IR (KBr)  $\nu_{\max}$  [ $\text{cm}^{-1}$ ]:

3408, 3095, 2922, 1673, 1611, 1601, 1547, 1474, 1383, 1310, 1228, 1184, 1159, 1109, 1052, 965, 823, 791, 771, 701, 665, 617, 598.

30

C. Synthesis of 4,4'-di(8-hydroxy-5-sulfo-7-quinolinecarboxyamido)stilbene-2,2'-disulfonic acid tetrasodium salt

5 To a solution of 4,4'-diaminostilbene-2'-disulfonic acid disodium salt (2.5 mmol, 1.036 g) in DMF (170 mL) a solution of 7-carboxy-8-hydroxyquinoline-5-sulfonic acid monosodium salt (5.0 mmol, 1.456 g) in DMF (650 mL) was added. To the mixture, 1-hydroxybenzotriazole (HBT) (15.0  
10 mmol, 2.027 g) was added followed by 1,3-dicyclohexylcarbodiimide (DCC) (20.0 mmol, 4.127 g). The reaction mixture was stirred at room temperature for 8 days. After that time the reaction solution was concentrated under reduced pressure. The precipitate was  
15 removed by filtration. The filtrate was evaporated in vacuo. Water was added to the residue and the mixture was evaporated under reduced pressure. Next the residue was extracted with chloroform (3x). Insoluble material was filtered off and treated with methanol, with its pH being  
20 adjusted to 8. A small amount of insoluble material was removed by filtration. The methanol solution was used for fractionation on Sephadex LH-20 column using methanol as the eluent. This technique allowed for the separation of pure diamide in 20% yield. The fractionation was repeated  
25 for non-homogeneous fractions containing the desired compound together with a small amount of monoamide. Finally, the pure diamide was separated from the reaction mixture in 25% yield. Its substituents and their position of attachment are shown in Table 1, under Compound 1.

30 m.p. >300°C  
<sup>1</sup>H NMR (D<sub>2</sub>O) δ [ppm]:  
8.75 (br d, J=8.1 Hz, 2H, H-a)  
8.67 (br d, J=3.9 Hz, 2H, H-c)  
8.54 (s, 2H, H-d)  
35 8.22 (d, J=2.1 Hz, 2H, H-e)  
7.89 (d, J=8.7 Hz, 2H, H-g)  
7.86 (s, 2H, H-h)  
7.78 (dd, J=8.4; 1.8 Hz, 2H, H-f)

7.56 (dd,  $J=8.1$ ; 4.2 Hz, 2H, H-b)

IR (KBr)  $\nu_{\max}$  [ $\text{cm}^{-1}$ ]:

3443, 2926, 2855, 1641, 1580, 1530, 1499, 1451, 1405, 1306, 1190, 1113, 1080, 1061, 1042, 1025, 761, 703, 624.

5 UV ( $\text{H}_2\text{O}$ )  $\nu_{\max}$  [nm]:

204, 244, 277, 367

Anal. ( $\text{C}_{34}\text{H}_{20}\text{N}_4\text{Na}_4\text{O}_{16}\text{S}_4 \times 9\text{H}_2\text{O}$ ) C, H, N, S

#### EXAMPLE 2

10 This Example illustrates the synthesis of the aroylaniline derivative, 4-amino-4'-(8-hydroxy-5-sulfo-7-quinolinecarboxyamido)stilbene-2,2'-disulfonic acid trisodium salt, Compound 2 as shown in Table 1. To a solution of 4,4'-diaminostilbene-2,2'-disulfonic acid  
15 disodium salt (0.5572 mmol, 0.234 g) in DMF (35 mL) a solution of 7-carboxy-8-hydroxyquinoline-5-sulfonic acid monosodium salt (1.114 mmol, 0.324 g) in DMF (65 mL) was added. To the mixture 1-hydroxybenzotriazole (HBT) (1.332 mmol, 0.180 g) and then 1,3-dicyclohexylcarbodiimide (DCC)  
20 (2.228 mmol, 0.360 g) were added. The reaction mixture was refluxed under nitrogen for 3 days. Then, insoluble material was filtered off to give 0.073 g of the product of 7-carboxy-8-hydroxyquinoline-5-sulfonic acid monosodium salt self-esterification. The filtrate was evaporated  
25 under reduced pressure. Water was added to the residue and insoluble material was filtered off. The filtrate was concentrated in vacuo and the pH of solution was adjusted to 9. The solvent was removed under reduced pressure. The residue was treated with methanol, insoluble material was  
30 removed by filtration, and the filtrate was concentrated under reduced pressure and fractionated on Sephadex LH 20 column using methanol as eluent. Fractionation was repeated for non-homogeneous fractions containing product. This technique allowed for the separation of the pure  
35 monoamide, Compound 2, in 42% yield (0.160 g).  
m.p.  $>300^\circ\text{C}$

$^1\text{H}$  NMR ( $\text{D}_2\text{O}$ )  $\delta$  [ppm]:

- 8.77 (br d,  $J=8.4$  Hz, 1H, H-a)  
8.67 (br d,  $J=4.2$  Hz, 1H, H-c)  
8.52 (s, 1H, H-d)  
5 8.16 (d,  $J=2.1$  Hz, 1H, H-e)  
7.76 (m, 4H, H-f, H-g, H-h, H-i)  
7.65 (d,  $J=8.4$  Hz, 1H, H-k)  
7.59 (dd,  $J=8.4$ ; 4.2 Hz, 1H, H-b)  
7.21 (d,  $J=2.4$  Hz, 1H, H-j)  
10 6.88 (dd,  $J=8.4$ ; 2.4 Hz, 1H, H-l)  
IR (KBr)  $\nu_{\text{max}}$  [ $\text{cm}^{-1}$ ]  
3444, 2956, 2925, 2854, 1635, 1605, 1577, 1526, 1498, 1456,  
1449, 1405, 1307, 1188, 1140, 1113, 1083, 1049, 1025, 797,  
764, 730, 706, 622.  
15 Anal. ( $\text{C}_{24}\text{H}_{16}\text{N}_3\text{Na}_3\text{O}_{11}\text{S}_3 \times 5\text{H}_2\text{O}$ ) C, H, N, S

### EXAMPLE 3

This Example illustrates the synthesis of 1,3-di[2,2'-disulfo-4(8-hydroxy-5-sulfo-7-quinoline)]stilbene urea, hexasodium salt, Compound 3 as shown in Table 1.

A solution of the monoamide from Example 2 above (0.07 mmol, 0.048 g) and triethylamine (0.2 mL) in DMF (50 mL) was added dropwise over 20 minutes to a solution of triphosgene (0.14 mmol, 0.042 g) in DMF (5 mL) stirred at 25 5-10°C. Then, the reaction mixture was stirred and heated at 120°C for 6 hours followed by evaporation under reduced pressure. The solid was treated with methanol, and the pH of the methanol solution was adjusted to 8. The material was filtered, and filtrate was concentrated and used for 30 fractionation on Sephadex LH-20 column, with the eluent being methanol. All fractions were analyzed by  $^1\text{H}$  NMR spectroscopy. This technique allowed for the obtaining of the desired product, Compound 3.

m.p. >300°C

35  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ )  $\delta$  [ppm]:

8.75 (br d,  $J=8.1$  Hz, 2H, H-a)



8.68 (br d, J=3.0 Hz, 2H, H-c)

8.53 (s, 2H, H-d)

8.20 (br s, 2H, H-e)

7.83, 7.59 (2 m, 16H, H-b, H-f - H-1)

5 IR (KBr)  $\nu_{\max}$  [cm<sup>-1</sup>]:

3439, 2961, 2926, 2850, 1735, 1660, 1460, 1397, 1193, 1139,  
842, 631

#### EXAMPLE 4

10 This Example illustrates the synthesis of the  
aroylaniline derivative, 4,4'-di(8-hydroxy-7-  
quinolinecarboxyamido)stilbene-2,2'-disulfonic acid  
disodium salt, Compound 4 as shown in Table 1.

#### 15 A. Carboxylation of 8-hydroxyquinoline

Potassium salt of 8-hydroxyquinoline (25 mmol, 4.58  
g), DMF (50 mL) and an excess of dry ice were heated in a  
pressure reactor at 120°C for 12 hours. After cooling,  
water (150 mL) was added to the reaction mixture. The  
20 insoluble material was filtered off, and the filtrate was  
evaporated under reduced pressure. To the residue, water  
was added again, and additional insoluble material was  
collected by filtration. The filtrate was acidified to pH  
4 with concentrated hydrochloric acid. The precipitate was  
25 filtered off, washed with water and dried to give 7-  
carboxy-8-hydroxyquinoline in 30.4% yield (1.44 g).

m.p. 264-266°C

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  [ppm]:

8.91 (dd, J=4.8; 1.5 Hz, 1H, H-a)

30 8.61 (dd, J=8.4; 1.5 Hz, 1H, H-c)

7.91 (d, J=8.4 Hz, 1H, H-e)

7.80 (dd, J=8.4; 4.5 Hz, 1H, H-b)

7.31 (d, J=8.7 Hz, 1H, H-d)

IR (KBr)  $\nu_{\max}$  [cm<sup>-1</sup>]

3429, 3081, 2932, 1644, 1589, 1549, 1480, 1458, 1420, 1383, 1333, 1203, 1150, 1114, 936, 874, 846, 830, 800, 763, 690, 644, 600.

5    B. Synthesis of 4,4'-di(8-hydroxy-7-quinolinecarboxyamido)stilbene-2,2'-disulfonic acid disodium salt

To a solution of 4,4'-diaminostilbene-2,2'-disulfonic acid disodium salt (2.5 mmol, 1.036 g) in DMF (170 mL) a solution of 8-hydroxyquinoline-7-carboxylic acid (5.0 mmol, 0.946 g) in DMF (140 mL) was added. To the mixture, 1-hydroxybenzotriazole (HBT) (6.0 mmol, 0.811 g) was added followed by 1,3-dicyclohexylcarbodiimide (DCC) (12.0 mmol, 2.476 g). The reaction mixture was stirred at room temperature for 6 days. Then, the solution was concentrated under reduced pressure. The precipitate was filtered off. The filtrate was evaporated in vacuo. To the residue, water and methanol (1:4) were added, and the pH was adjusted to 8. The insoluble material was filtered off, and the filtrate was evaporated under reduced pressure. The residue was extracted with chloroform (3x). The solid was collected by filtration, and treated with methanol. Insoluble material was filtered off and dried to give 0.8 g of pure, desired diamide. The methanol solution was used for fractionation on a Sephadex LH-20 column, with the eluent being methanol. This method allowed for the separation of an additional 0.59 g pure diamide (yield 73.5%) and traces of monoamide.

30    m.p. >300°C

<sup>1</sup>H NMR (D<sub>2</sub>O) δ [ppm]:

8.55 (d, J=3.0 Hz, 2H, H-a)

8.16 (d, J=2.1 Hz, 2H, H-f)

7.98 (br d, J=8.1 Hz, 2H, H-c)

35    7.78 (m, 6H, H-e, H-h, H-i)

7.61 (dd, J=8.4; 2.1 Hz, 2H, H-g)

7.33 (dd, J=8.4; 4.2 Hz, 2H, H-b)

6.79 (d, J=8.7 Hz, 2H, H-d)

IR (KBr)  $\nu_{\max}$  [ $\text{cm}^{-1}$ ]

3452, 2958, 2925, 2854, 1635, 1590, 1578, 1527, 1503, 1491,  
1453, 1404, 1304, 1218, 1186, 1134, 1109, 1081, 1024, 825,  
796, 779, 738, 711, 630.

5 Anal. ( $\text{C}_{24}\text{H}_{22}\text{N}_4\text{Na}_2\text{O}_{16}\text{S}_2 \times 6\text{H}_2\text{O}$ ) C, H, N, S

#### Examples 5-12

Following the procedures set forth in Example 1 and  
acylating amine (II) with the appropriate, activated acid  
10 (III), the quinoline compounds of Compounds 5-12 as shown  
in Table 1 are obtained.

#### Example 13

This Example illustrates the synthesis 4,4'-di[5-(2-  
15 carboxyethylthiomethyl) - 8 - hydroxy - 7 -  
quinolinecarboxyamido]stilbene-2,2'-disulfonic acid  
tetrasodium salt, Compound 13 as shown in Table 1, by 2-  
carboxyethylthiomethylation of the diamide of Example 4.

The above diamide, Compound 4, (0.3 mmol, 0.227 g), 3-  
20 mercaptopropionic acid (0.63 mmol, 0.0669 g), formalin  
(0.66 mmol, 0.0198 g), and an aqueous solution of NaOH  
(1.23 mmol, 0.0492 g/3 mL of  $\text{H}_2\text{O}$ ) were refluxed under  
nitrogen for 1.5 hours. Then, more formalin (0.33 mmol,  
0.01 g) was added, and refluxing was continued for 30  
25 minutes. After cooling, a very small amount of insoluble  
material was filtered off. The filtrate was evaporated  
under reduced pressure. The residue was treated with  
methanol, and the pH of the solution was adjusted to 8. A  
methanol soluble material was used for fractionation on a  
30 Sephadex LH-20 column, with the eluent being methanol.  
This technique allowed the separation of the pure product,  
Compound 13, in 51.5% yield (0.16 g).

m.p.  $>300^\circ\text{C}$

$^1\text{H}$  NMR ( $\text{D}_2\text{O}$ )  $\delta$  [ppm]:

35 8.55 (br d,  $J=3.3$  Hz, 2H, H-a)

8.22 (br s, 2H, H-e)

8.12 (br d, J=8.4 Hz, 2H, H-c)  
7.69 (s, 2H, H-h)  
7.54 (d, J=8.7 Hz, 2H, H-g)  
7.46 (s, 2H, H-d)  
5 7.39 (m, 4H, H-b, H-f)  
3.60 (s, 4H, H-i)  
2.53 (t, J=7.2 Hz, 4H, H-j)  
2.31 (t, J=7.2 Hz, 4H, H-k)  
IR (KBr)  $\nu_{\max}$  [cm<sup>-1</sup>]  
10 3449, 2924, 2860, 1636, 1576, 1544, 1527, 1504, 1491, 1458,  
1399 1310, 1217, 1181, 1141, 1080, 1024, 824, 769, 735,  
707, 629.

#### Examples 14-18

15 Following the procedure of Example 13 and reacting the  
diamide of Example 4 having the appropriate X-substituents  
with formaldehyde and the appropriate mercaptoalkanoic  
acid, mercaptoalkanoic acid ester or mercaptoalkanoic acid  
amide, the quinoline compounds of Compounds 14-18 were  
20 obtained.

#### Example 19

This Example illustrates the synthesis of the  
aroylaniline derivative, 4,4'-di(8-hydroxy-7-  
25 quinolinethioamido)stilbene-2,2'-disulfonic acid disodium  
salt, Compound 19 shown in Table 1.

#### A. Synthesis of 7-thiocarboxy-8-hydroxyquinoline

A mixture of 8-hydroxyquinoline (0.055 mol, 8.0 g) and  
30 O-ethylxanthic acid potassium salt (0.056 mol, 9.0 g) in  
abs. EtOH (20 mL) was placed in a pressure reactor and  
heated at 160°C for 40 hours. After cooling, the reaction  
mixture was washed with a small amount of EtOH. The  
insoluble material was filt red off and treated with  
35 aqueous hydrochloric acid. The insoluble thioacid was  
separated by filtration and purified by heating with barium

carbonate in water. The mixture was filtered off, the filtrate was acidified with hydrochloric acid, and precipitate was collected by filtration to give 7-thiocarboxy-8-hydroxyquinoline (2.61 g) in 18.4% yield.

5 m.p. 174-175°C

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ [ppm]:

9.06 (m, 2H, H-a, H-c)

9.00 (d, J=9.0 Hz, 1H, H-e)

8.06 (dd, J=8.4; 5.4 Hz, 1H, H-b)

10 7.58 (d, J=9.0 Hz, 1H, H-d)

IR (KBr) ν<sub>max</sub> [cm<sup>-1</sup>]:

3471, 3090, 3064, 3038, 2965, 2922, 1630, 1606, 1585, 1541,  
1499, 1474, 1417, 1402, 1373, 1319, 1287, 1242, 1225, 1197,  
1176, 1138, 1111, 1049, 1013, 997, 982, 972, 874, 824, 811,  
15 794, 780, 766, 718, 670.

B. Synthesis of 4,4'-di(8-hydroxy-7-quinolinethioamido)stilbene-2,2'-disulfonic acid disodium salt

20

To a solution of 4,4'-diaminostilbene-2,2'-disulfonic acid disodium salt (1.0 mmol, 0.414 g) in DMF (70 mL) a solution of 8-hydroxyquinoline-7-thiocarboxylic acid (2.0 mmol, 0.443 g) in DMF (330 mL) was added. To the mixture, 25 1-hydroxybenzotriazole (HBT) (2.4 mmol, 0.324 g) was added followed by 1,3-dicyclohexylcarbodiimide (DCC) (4.8 mmol, 0.990 g). The reaction mixture was stirred at room temperature for 6 days. Then, the solution was evaporated under reduced pressure. To the residue, water and methanol 30 (1:4) were added, and the pH was adjusted to 8. The insoluble material was filtered off, and the filtrate was evaporated under reduced pressure. The residue was extracted with chloroform (3x). The solid was collected by filtration and treated with methanol. The methanol 35 solution was used for fractionation on a Sephadex LH-20 column, with the eluent being methanol. The fractions containing the pure, desired product of Compound 19 were combined and evaporated under reduced pressure (yield 17%).

m.p. >300°C

<sup>1</sup>H NMR (D<sub>2</sub>O) δ [ppm]:

8.55 (dd, J=4.2; 1.5 Hz, 2H, H-a)

8.16 (d, J=2.1 Hz, 2H, H-e)

5 7.98 (dd, J=8.4; 1.5 Hz, 2H, H-c)

7.78 (m, 6H, H-e, H-h, H-i)

7.61 (dd, J=8.4; 2.1 Hz, 2H, H-g)

7.33 (dd, J=8.4; 4.5 Hz, 2H, H-b)

6.79 (d, J=8.7 Hz, 2H, H-d)

10 IR (KBr) ν<sub>max</sub> [cm<sup>-1</sup>]:

3467, 2923, 2855, 1635, 1591, 1527, 1502, 1490, 1452, 1403,  
1303, 1214, 1183, 1132, 1109, 1079, 1023, 891, 824, 796,  
779, 737, 711, 629.

15 Examples 20-23

Following the procedure set forth in Example 19 and reacting the appropriate dithioamides with formaldehyde and the appropriate mercaptoalkanoic acid, mercaptoalkanoic acid ester or mercaptoalkanoic acid amide, the quinoline  
20 Compounds 20-23 were obtained.

Example 24

This Example illustrates the synthesis of the aroylaniline derivative, 4,4'-di[4-(2-  
25 carboxyethylthiomethyl)-1-hydroxy-2-naphthamido]stilbene-2-  
2'-disulfonic acid tetrasodium salt, Compound 24 as shown in Table 1.

30 A. Synthesis of 4,4'-di(1-hydroxy-2-naphthamido)stilbene-2-2'-disulfonic acid disodium salt

To a solution of 4,4'-diaminostilbene-2,2'-disulfonic acid disodium salt (2.0 mmol, 0.829 g) in DMF (140 mL) a solution of 1-hydroxy-2-naphthoic acid (4.0 mmol, 0.753 g)  
35 in DMF (60 mL) was added. To the mixture, 1-hydroxybenzotriazole (HBT) (5.0 mmol, 0.676 g) was added followed by 1,3-dicyclohexylcarbodiimide (DCC) (10 mmol,

2.060 g). The reaction mixture was stirred at room temperature for 7 days. Then, the precipitate was filtered off, and the filtrate was evaporated under reduced pressure. To the residue, water was added, and the solvent  
5 was removed in vacuo. Next, the residue was extracted with chloroform (3x). The solid was collected by filtration and treated with methanol, and the pH of the solution was adjusted to 8. The methanol solution was concentrated under reduced pressure and used for fractionation on a  
10 Sephadex LH-20 column, with the eluent being methanol. This technique allowed for the separation of the pure desired diamide in 49% yield (0.74 g).

m.p. >300°C

<sup>1</sup>H NMR (D<sub>2</sub>O) δ [ppm]:

- 15 8.26 (m, 2H, H-a)  
8.10 (d, J=2.1 Hz, 2H, H-g)  
7.75 (m, 4H, H-f, H-j)  
7.70 (d, J=8.7 Hz, 2H, H-i)  
7.57 (m, 2H, H-d)  
20 7.52 (dd, J=8.7; 2.1 Hz, 2H, H-h)  
7.34 (m, 4H, H-b, H-c)  
6.90 (d, J=9.0 Hz, 2H, H-e)

IR (KBr) ν<sub>max</sub> [cm<sup>-1</sup>]:

- 25 3439, 2924, 2854, 1631, 1594, 1525, 1505, 1469, 1392, 1359,  
1300, 1247, 1208, 1082, 1026, 809, 790, 761, 726, 707, 632,  
542.

- 30 B. 4,4'-di[4-(2-carboxyethylthiomethyl)-1-hydroxy-2-naphthamido]stilbene-2,2'-disulfonic acid tetrasodium salt

The above diamide (0.3 mmol, 0.226 g), 3-mercaptopropionic acid (0.63 mmol, 0.0669 g), formalin  
35 (0.66 mmol, 0.0198 g), and aqueous solution of NaOH (1.23 mmol, 0.0492 g/3 mL of H<sub>2</sub>O) were refluxed under nitrogen for 1.5 hours. Then, more formalin (0.33 mmol, 0.01 g) was added and refluxing was continued for a further 30 minutes. After cooling, a small amount of insoluble material was  
40 filtered off. The filtrate was evaporated under reduced pressure. The residue was treated with methanol, and the pH of the solution was adjusted to 8. Methanol soluble material was used for fractionation on a Sephadex LH-20

column, with the eluent being methanol. This method allowed for the separation of the pure Compound 24 in 41.9% yield (0.13 g).

m.p. >300°C

5  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ )  $\delta$  [ppm]:

8.32 (br d,  $J=7.8$  Hz, 2H, H-a)

8.17 (d,  $J=2.4$  Hz, 2H, H-f)

7.86 (br d,  $J=8.1$  Hz, 2H, H-d)

7.75 (m, 4H, H-h, H-i)

10 7.63 (m, 4H, H-e, H-g)

7.53 (m, 2H, H-b)

7.41 (m, 2H, H-c)

3.94 (s, 2H, H-j)

2.62 (t,  $J=7.2$  Hz, 2H, H-k)

15 2.38 (t,  $J=7.2$  Hz, 2H, H-l)

IR (KBr)  $\nu_{\text{max}}$  [ $\text{cm}^{-1}$ ]:

3558, 3380, 2924, 2854, 1632, 1576, 1524, 1490, 1427, 1399, 1308, 1193, 1140, 1081, 1026, 764, 727, 708, 627.

20

#### Examples 25-28

Following the procedure of Example 24 and reacting the appropriate diamides with formaldehyde, and the appropriate mercaptoalkanoic acid, mercaptoalkanoic acid ester or  
25 mercaptoalkanoic acid amide, the naphthalene Compounds 25-28 are obtained.

#### Example 29

This Example illustrates the synthesis of the  
30 aroylaniline derivative, 4,4'-di(2-carboxy-3,6-dihydroxybenzamido)stilbene-2,2'-disulfonic acid disodium salt, Compound 29 as shown in Table 1.

#### A. Hydrolysis of 2,3-dicyanohydroquinone

35 2,3-Dicyanohydroquinone (29 mmol, 4.7 g) was mixed into 50% KOH (40 mL). The mixture was heated under nitrogen to reflux for 1.5 hours, then cooled to room temperature. The mixture was neutralized with aqueous  $\text{H}_2\text{SO}_4$  and then extracted with ethyl acetate (5 x 100 mL).  
40 Evaporation of the solvent and then careful crystallization from water gave the diacid, mp 216-219°C (dec). Yield 3.0



g (52%). <sup>1</sup>H-NMR (DMSO, ppm) 6.8 (s, 1H). FT-IR (KBr, cm<sup>-1</sup>) 3249, 1706, 1667, 1450, 1192.

#### B. Anhydride Formation

5       The above diacid (1.0 g) was heated in a vacuum sublimation apparatus to 230°C using tap water as the cooling source and an aspirator vacuum. The sublimed material was a sulfur yellow solid which was recrystallized from benzene. Yield was virtually quantitative. <sup>1</sup>H-NMR  
10       (DMSO, ppm) 7.2 (s, 1H). FT-IR (KBr, cm<sup>-1</sup>) 3600, 1827, 1751, 1490, 1385, 1306, 1197, 894. EIMS: 180 (100), 162 (75), 136 (21), 134 (519).

#### C. Condensation

15       4,4'-Diaminostibene-2,2' disulfonic acid disodium salt (0.4 mmol, 0.1 g) was stirred in DMF (20 mL) containing 4 Å molecular sieves overnight. Afterwards, the anhydride (0.16 g) from Part B above was added, and the mixture was stirred at 70°C for 2 hours. After cooling and filtering,  
20       acetone was added, and the solid was collected and washed with copious amounts of acetone. After addition to 0.044 g of NaHCO<sub>3</sub> in 10 mL of water, the crude material was lyophilized to give the diamide of Compound 29, as shown in Table 1, as a light brown powder. <sup>1</sup>H-NMR (DMSO, ppm) 9.82  
25       (s, 1H), 8.18 (d, 1H, J=2Hz), 7.99 (s, 1H), 7.63 (dd, 1H, J=2, 8Hz), 7.50 (d, 1H, J=8Hz), 6.72 (d, 1H, J=8Hz), 6.55 (d, 1H, J=8Hz). FT-IR (KBr, cm<sup>-1</sup>) 3400 (B), 1631, 1589, 1384, 1083, 1026, 632.

#### 30       Examples 30-31

Following the procedure set forth in Example 29 and reacting the appropriate aniline (II) with the anhydride of Example 29, Part B, Compounds 30-31 were obtained.

Example 32

This Example illustrates the synthesis of the aroylaniline derivative, N,N'-Bis(4-amino-2-sulfobenzoyl)-4,4'-diaminostilbene-2,2'-disulfonic acid tetrasodium salt, Compound 32 as shown in Table 1.

A. N,N'-Bis(4-nitro-2-sulfobenzoyl)-4,4'-diaminostilbene-2,2'-disulfonic Acid Tetrasodium Salt

4-Nitro-2-methylsulfonate benzoyl chloride (0.6 g, 2.0 mmol) was dissolved in ice-cold dry acetonitrile (15 ml). This solution was added dropwise to a solution of 4,4'-diaminostilbene-2,2'-disulfonic disodium salt (504 mg, 1.0 mmol) in water (25 ml) at 0°C. The reaction mixture was carefully kept at pH 7 with the addition of a dilute sodium hydroxide solution and stirred for two hours in an ice bath. A small amount of insoluble materials was filtered off, and the clear yellow solution was freeze-dried to give a mixture containing mainly the product. Gel filtration on Sephadex G-25 with water yielded 0.95 g (96%) of the pure product. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ [ppm]: 11.0 (1H, s), 8.61 (1H, d, J=2.3 Hz), 8.31 (1H, dd, J<sub>1</sub>=8.4 Hz, J<sub>2</sub>=2.3 Hz), 8.05 (1H, s), 8.03 (1H, d, J=2.2 Hz), 7.96 (1H, d, J=8.3 Hz), 7.78 (1H, dd, J<sub>1</sub>=8.4 Hz, J<sub>2</sub>=2.2 Hz), 7.61 (1H, d, J=8.7 Hz).

B. N,N'-Bis(4-amino-2-sulfobenzoyl)-4,4'-diaminostilbene-2,2'-disulfonic Acid Tetrasodium Salt

To a sodium hydrogen sulfide solution the product from Part A (0.90 g) was added and the solution was refluxed for 45 minutes. Hydrochloric acid was added to the solution until the pH reached 6, and then the solvent was evaporated under reduced pressure. Water (200 mL) was added, and the resulting precipitate was collected by filtration and washed with water to give 0.45 g (53%) of the title compound. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ [ppm]: 11.7 (1H, s), 8.00 (1H, s), 7.93 (1H, d, J=2.2 Hz), 7.80 (1H, dd, J<sub>1</sub>=8.6 Hz, J<sub>2</sub>=2.1 Hz), 7.58 (1H, d, J=8.4 Hz), 7.54 (1H, d, J=8.8 Hz),

7.18 (1H, d, J=2.3 Hz), 6.55 (1H, dd, J<sub>1</sub>=8.4 Hz, J<sub>2</sub>=2.3 Hz), 5.72 (2H, broad).

### C. Coupling

5       The product from Part B, (240 mg, 0.17 mmol) and sodium nitrite (23 mg, 0.33 mmol) were dissolved in water (50 mL). The solution was then poured into a mixture of ice and concentrated hydrochloric acid (0.1 mL, 0.1 g, 0.33 mmol) and stirred for 30 minutes. A second solution was  
10       made from 8-hydroxyquinoline-5-sulfonic acid monohydrate (81 mg, 0.33 mmol) and sodium carbonate (38 mg, 0.33 mmol) in water (20 mL). While cooled in an ice bath, the two solutions were mixed and stirred for an hour. The mixture was adjusted to pH 8 by the addition of a small amount of  
15       sodium hydroxide solution. The coupling product (240 mg, 62%) was separated by salting out with sodium acetate. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ [ppm]: 11.6 (1H, s), 8.90 (1H, d, J=8.2 Hz), 8.54 (1H, d, J=2.2 Hz), 8.33 (1H, s), 8.25 (1H, s, J=small), 8.05 (1H, s), 8.03 (1H, d, J=small), 7.86 (1H, d, J=8.3 Hz), 7.84 (1H, dd, J<sub>1</sub>=7.7 Hz, J<sub>2</sub>=small), 7.76 (1H, d, J=7.7 Hz), 7.60 (1H, dd, J<sub>1</sub>=8.1 Hz, J<sub>2</sub>=small), 7.44 (1H, dd, J<sub>1</sub>=8.2 Hz, J<sub>2</sub>=4.3 Hz). The compound's substituents and their position of attachment are shown in Table 1.

### 25       Examples 33-34

Following the procedure set forth in Example 32 and coupling the appropriate diazonium salt derived from the appropriate aniline (II) with 8-hydroxyquinoline-5-sulfonic acid, Compounds 33-34 are obtained.

30

### Example 35

The aroylaniline derivatives of the present invention were subjected to *in vitro* screening to determine their anti-HIV activity.

35       The National Cancer Institute's procedure for testing for agents active against Human Immunodeficiency Virus

(HIV) (Weislow, O.W. et al., "New soluble-formazan assay for HIV-1 cytopathic effects: application to high-flux screening of synthetic and natural products for AIDS-antiviral activity," J. Natl. Cancer Inst., 81, 577-586 (1989)) is designed to detect agents acting at any stage of the virus reproductive cycle. The assay includes the killing of T4 lymphocytes by HIV. Pursuant to the test, small amounts of HIV are added to cells, and a complete cycle of virus reproduction is necessary to obtain the required cell killing. Agents that interact with virions, cells, or virus gene-products to interfere with viral activities will protect cells from cytolysis. The system is thus designed to detect anti-HIV activity. However, compounds that are chemically labile or are rapidly metabolized in the culture conditions may not show activity in this screen. All tests are compared with at least one positive (e.g., AZT-treated) control done at the same time under identical conditions.

The Procedure:

1. Candidate agent is dissolved in dimethyl sulfoxide (unless otherwise instructed) and then diluted 1:100 in cell culture medium before preparing serial half-log<sub>10</sub> dilutions which are then to 96-well microfilter plates. T4 lymphocytes (CEM cell line) are added, and after a brief interval HIV-1 is added to all wells, resulting in a 1:200 final dilution of the compound. Uninfected cells with the compound serve as a toxicity control, and infected and uninfected cells without the compound serve as basic controls.
2. Cultures are incubated at 37°C in a 5% carbon dioxide atmosphere for 6 days.
3. The tetrazolium salt, XTT, is added to all wells, and cultures are incubated to allow formazan color development by viable cells.
4. Individual wells are analyzed spectrophotometrically to quantitate formazan production, and in addition are

viewed microscopically for detection of viable cells and confirmation of protective activity.

5. Drug-tested virus-infected cells are compared with drug-treated noninfected cells and with other appropriate controls (untreated infected and untreated noninfected cells, drug-containing wells without cells, etc.) on the same plate.

6. Data are reviewed in comparison with other tests done at the same time and a determination about activity is made.

The aroylaniline derivatives of the present invention were tested for anti-HIV activity according to the above procedure. Figures 1-6 represent test results of the aroylaniline derivatives of Compounds 1, 3, 13, 14, 29 and 32.

Figures 1 to 6 display a plot of the  $\log_{10}$  of the concentrations (as  $\mu\text{g/mL}$  or molar as shown) of the aroylaniline derivatives, Compounds 1, 3, 13, 14, 29 and 32, against the measured test values expressed as a percentage of the uninfected, untreated control values. The solid line depicts the percentage of surviving HIV-infected cells treated with sample (at the indicated concentration) relative to uninfected, untreated controls. Thus, the solid line expresses the anti-HIV activity of the compound. The dashed line depicts the percentage of surviving uninfected cells treated with the compound relative to the same uninfected, untreated controls. Thus, the dashed line expresses the growth inhibitory properties of the compound. The viral cytopathic effect of the compound is indicated by a dotted reference line. The dotted reference line shows the extent of destruction of cells by the virus in the absence of treatment and is used as a quality control parameter. Survival values of this parameter less than 50% are considered acceptable in the current protocol. Approximate values for 50% effective concentration ( $\text{EC}_{50}$ ) have been calculated for each test and

are set forth for these Compounds in Tables 2-7, respectively.

Table 2 represents the *in vitro* anti-HIV data for the graph shown in Figure 1.

5

TABLE 2

Summary		Dose (Molar)	Percent of Protec- tion	Percent of Control	
Index	Concentration			Infected	Un- infected
IC50(Molar)	$1.20 \times 10^{-4}$	$6.35 \times 10^{-8}$	-5.09	20.13	97.61
EC50(Molar)	$1.44 \times 10^{-6}$	$2.00 \times 10^{-7}$	5.09	27.87	102.04
TI50(IC/EC)	$8.28 \times 10^{+1}$	$6.34 \times 10^{-7}$	3.41	26.59	105.10
Conclusion		$2.00 \times 10^{-6}$	68.63	76.16	106.46
CONFIRMED ACTIVE		$6.33 \times 10^{-6}$	121.16	116.08	117.18
		$2.00 \times 10^{-5}$	107.11	105.40	108.84
		$6.32 \times 10^{-5}$	85.61	89.06	104.08
		$2.00 \times 10^{-4}$	-25.97	4.26	6.47

15

Table 3 represents the *in vitro* anti-HIV data for the graph shown in Figure 2.

TABLE 3

Summary		Dose (Molar)	Percent of Protec- tion	Percent of Control	
Index	Concentration			Infected	Un- infected
IC50(Molar)	$>5.00 \times 10^{-5}$	$1.58 \times 10^{-8}$	0.19	4.18	110.80
EC50(Molar)	$4.06 \times 10^{-6}$	$5.02 \times 10^{-8}$	0.51	4.49	95.85
TI50(IC/EC)	$>1.23 \times 10^{-1}$	$1.58 \times 10^{-7}$	-1.29	2.76	94.42
Conclusion		$5.01 \times 10^{-7}$	2.00	5.92	97.72
CONFIRMED ACTIVE		$1.58 \times 10^{-6}$	0.19	4.18	95.67
		$5.00 \times 10^{-6}$	61.15	62.70	97.00
		$1.58 \times 10^{-5}$	92.90	93.18	103.32
		$5.00 \times 10^{-5}$	90.76	91.13	97.98

25

Table 4 represents the *in vitro* anti-HIV data for the graph shown in Figure 3.

TABLE 4

Summary		Dose (Molar)	Percent of Protec- tion	Percent of Control	
Index	Concentration			Infected	Un- infected
IC50(Molar)	$2.90 \times 10^{-5}$	$6.35 \times 10^{-8}$	15.57	35.83	86.36
EC50(Molar)	$1.05 \times 10^{-6}$	$2.00 \times 10^{-7}$	-17.92	10.38	90.55
TI50(IC/EC)	$2.78 \times 10^{+1}$	$6.34 \times 10^{-7}$	24.57	42.67	92.22
Conclusion		$2.00 \times 10^{-6}$	84.38	88.13	110.93
CONFIRMED ACTIVE		$6.33 \times 10^{-6}$	133.79	125.68	117.81
		$2.00 \times 10^{-5}$	67.54	75.33	73.70
		$6.32 \times 10^{-5}$	29.43	1.63	0.37
		$2.00 \times 10^{-4}$	-28.03	2.70	2.05

Table 5 represents the *in vitro* anti-HIV data for the graph shown in Figure 4.

TABLE 5

Summary		Dose (Molar)	Percent of Protec- tion	Percent of Control	
Index	Concentration			Infected	Un- infected
IC50(Molar)	$1.65 \times 10^{-5}$	$6.35 \times 10^{-6}$	16.02	18.54	80.16
EC50(Molar)	$8.05 \times 10^{-7}$	$2.00 \times 10^{-7}$	17.57	20.04	88.92
TI50(IC/EC)	$2.05 \times 10^{+2}$	$6.34 \times 10^{-7}$	24.99	27.24	121.68
Conclusion		$2.00 \times 10^{-6}$	146.91	145.50	205.56
CONFIRMED ACTIVE		$6.33 \times 10^{-6}$	150.93	149.40	220.68
		$2.00 \times 10^{-5}$	16.58	19.08	15.48
		$6.32 \times 10^{-5}$	6.56	9.36	2.52
		$2.00 \times 10^{-4}$	27.65	29.82	-0.84

Tabl 6 repr sents the *in vitro* anti-HIV data for the graph shown in Figure 5.

TABLE 6

Index	Summary		Dose (Molar)	Percent of Protection	Percent of Control	
	Concentration				Infected	Un-infected
IC50(Molar)	$>2.20 \times 10^{-4}$		$6.35 \times 10^{-8}$	-3.31	3.92	93.21
EC50(Molar)	$1.25 \times 10^{-5}$		$2.00 \times 10^{-7}$	-1.09	5.99	93.77
TI50(IC/EC)	$>1.61 \times 10^{+1}$		$6.34 \times 10^{-7}$	-0.52	6.52	92.80
Conclusion			$2.00 \times 10^{-6}$	0.58	7.54	95.15
CONFIRMED ACTIVE			$6.33 \times 10^{-6}$	2.47	9.30	96.67
			$2.00 \times 10^{-5}$	83.30	84.47	99.02
			$6.32 \times 10^{-5}$	92.72	93.23	100.14
			$2.00 \times 10^{-4}$	79.66	81.08	92.95

Table 7 represents the *in vitro* anti-HIV data for the graph shown in Figure 6.

TABLE 7

Index	Summary		Dose (Molar)	Percent of Protection	Percent of Control	
	Concentration				Infected	Un-infected
IC50(Molar)	$1.51 \times 10^{-4}$		$6.36 \times 10^{-8}$	-2.32	17.12	83.94
EC50(Molar)	$1.08 \times 10^{-6}$		$2.01 \times 10^{-7}$	-6.05	14.10	82.40
TI50(IC/EC)	$1.40 \times 10^{+2}$		$6.35 \times 10^{-7}$	9.69	26.85	95.11
Conclusion			$2.01 \times 10^{-6}$	97.49	97.97	98.97
CONFIRMED ACTIVE			$6.34 \times 10^{-6}$	106.98	105.65	94.34
			$2.00 \times 10^{-5}$	115.65	112.68	98.84
			$6.33 \times 10^{-5}$	118.99	115.38	95.50
			$2.00 \times 10^{-4}$	52.90	61.85	35.45

As illustrated in Figures 1-6 and Tables 2-7, the aroylaniline derivatives of the present invention are effective in protecting cells from cytolysis. Their therapeutic index,  $TI_{50}$  (IC/EC) is indicated on their graphs and in their respective tables.

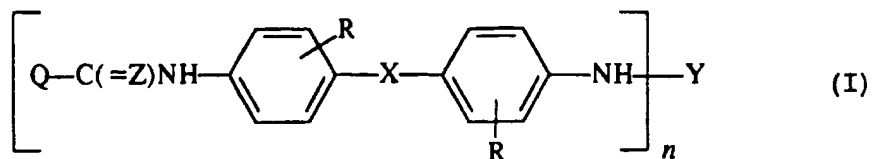


All references cited herein are incorporated in their entireties by reference.

While this invention has been described with emphasis upon preferred embodiments, it will be obvious to those skilled in the art that variations in the preferred compounds and methods may be used and that it is intended that the invention may be practiced otherwise than as specifically described herein. Accordingly, this invention includes all modifications encompassed within the spirit and scope of the invention as defined by the following claims.

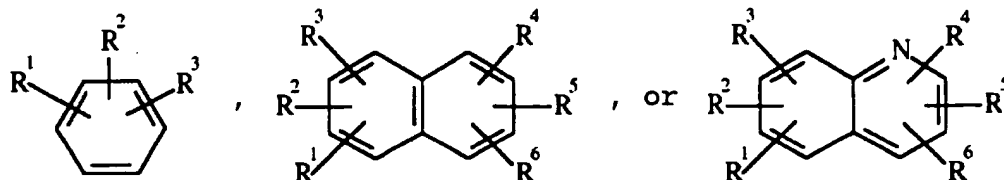
## WHAT IS CLAIMED IS:

1. An aroylaniline derivative having the formula:



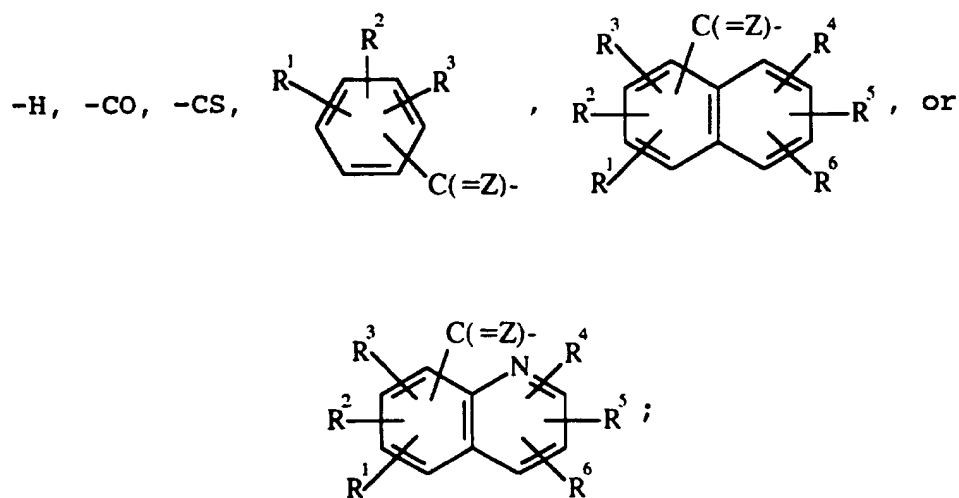
- 5 wherein R is -H, -SO<sub>3</sub>·E, -CO<sub>2</sub>·E, -PO(O)<sub>2</sub>·2E, -NO<sub>2</sub>, or a halogen;

Q is



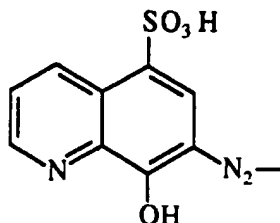
10

Y is



49

- $R^1, R^2, R^3, R^4, R^5$  and  $R^6$  are each independently  $-H$ ,  $-OH$ , a halogen,  $-NH_2$ ,  $-NHCH_3$ ,  $-N(CH_3)_2$ ,  $-CHO$ ,  $CH_3O-$ ,  $-SO_3 \cdot E$ ,  $-NO_2$ ,  $-CO_2 \cdot E$ ,  $-PO(O)_2 \cdot 2E$ ,  $-CH_2OH$ ,  $-CH_2SCH_2COR^7$ ,  $-CH_2S-$   
 5  $C_2H_4COR^7$ ,  $-CH_2S-CH(CO_2H)CH_2COR^7$ ,  $-CH_2S-CH(CO_2H)CH_2CO_2R^7$ ,  $-CH_2SCH_2CO_2R^7$ ,  $-CH_2SC_2H_4CO_2R^7$  or

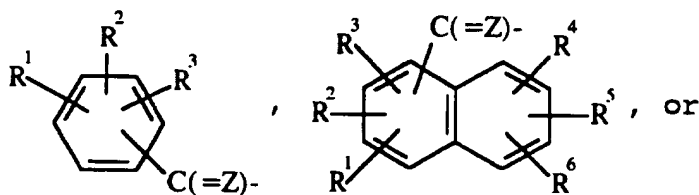


- 10 wherein  $E$  represents  $H$ , lithium, sodium, potassium, ammonium, magnesium,  $(HOC_2H_4)_3NH$ ,  $(CH_3)_3NH$ ,  $CH_3NH(C_2H_4OH)_2$ ,  $(CH_3)_4N-$ , or  $HN$ -methylglucamine and wherein  $R^7$  is  $H$ , methyl, ethyl, benzyl,  $-NH_2$ ,  $-NHCH_3$ , or  $-NHC_2H_5$ ;

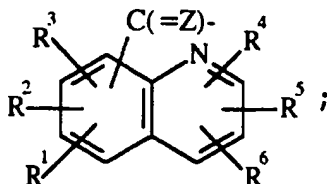
- $X$  is  $-CHCH-$ ,  $-CH_2CH_2-$ ,  $-CONH-$ ,  $-NHCONH-$ ,  $-NHCSNH-$ ,  $-S-$ ,  
 15  $-SO-$ ,  $-SO_2-$ ,  $-NH-$ ,  $-CH_2-$  or cyclopropyl;

$Z$  is oxygen or sulfur; and

$n$  is 1 or 2, but when  $n$  is 1,  $Y$  is  $H$ ,



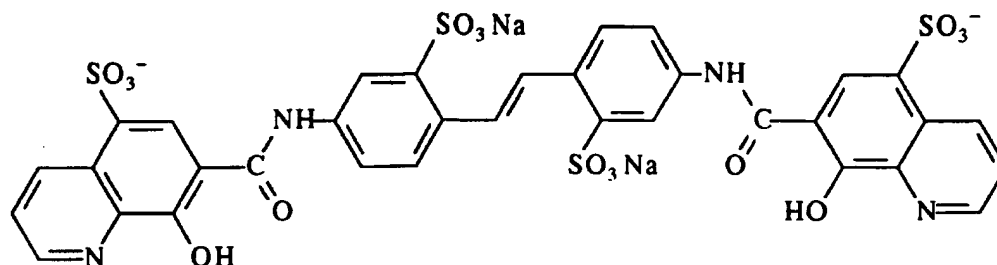
20



and when  $n$  is 2,  $Y$  is  $-CO$  or  $-CS$ .

50

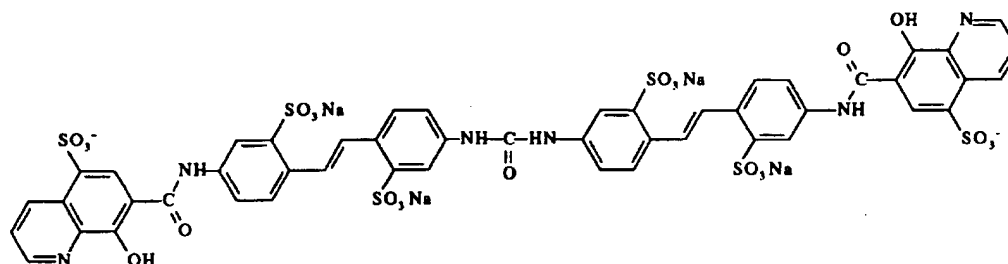
2. The aroylaniline derivative of claim 1 wherein the derivative has the formula:



5

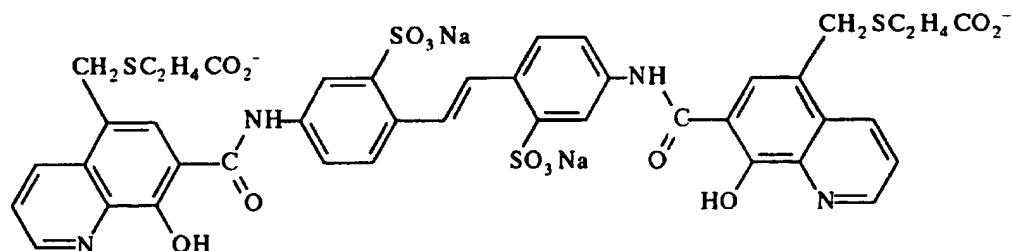
3. The aroylaniline derivative of claim 1 wherein the derivative has the formula:

10



15

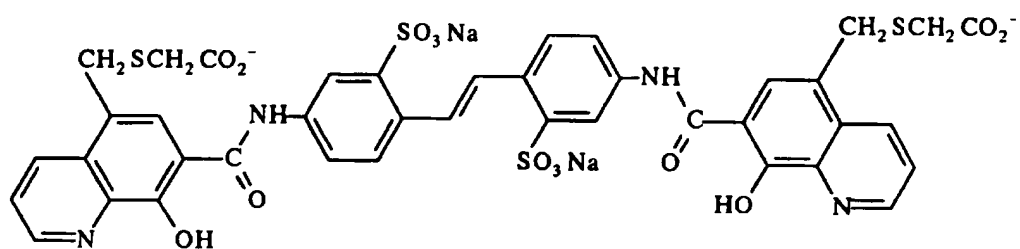
4. The aroylaniline derivative of claim 1 wherein the derivative has the formula:



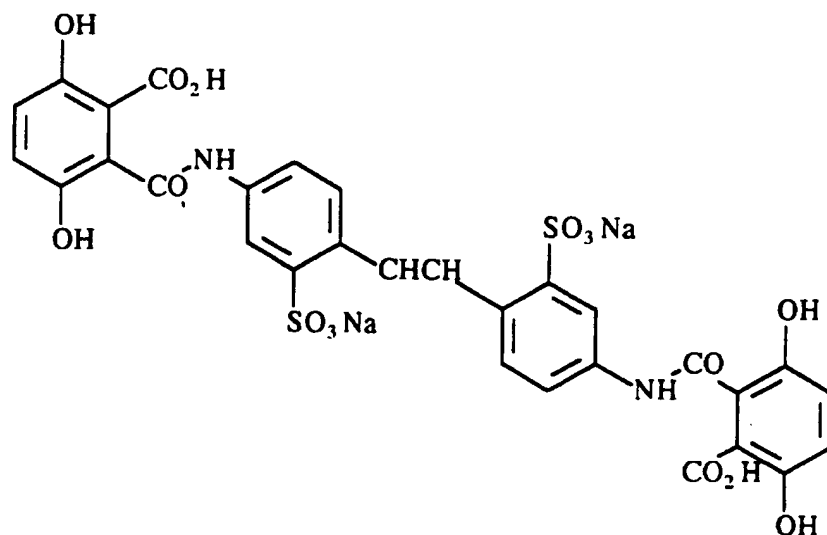
20

5. The aroylaniline derivative of claim 1 wherein the derivative has the formula:

51

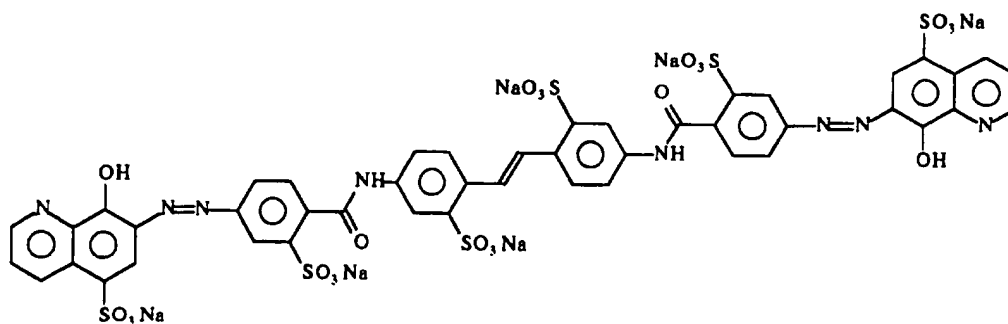


- 5            6. The aroylaniline derivative of claim 1 wherein the derivative has the formula:



10

7. The aroylaniline derivative of claim 1 wherein the derivative has the formula:



15

8. A pharmaceutical composition comprising the aroylaniline derivative of claim 1 and a pharmaceutically acceptable carrier.

5        9. A pharmaceutical composition comprising the aroylaniline derivative of claim 2 and a pharmaceutically acceptable carrier.

10       10. A pharmaceutical composition comprising the aroylaniline derivative of claim 3 and a pharmaceutically acceptable carrier.

15       11. A pharmaceutical composition comprising the aroylaniline derivative of claim 4 and a pharmaceutically acceptable carrier.

20       12. A pharmaceutical composition comprising the aroylaniline derivative of claim 5 and a pharmaceutically acceptable carrier.

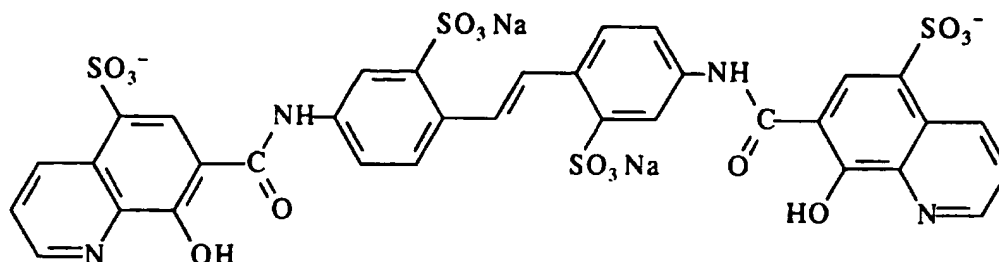
20       13. A pharmaceutical composition comprising the aroylaniline derivative of claim 6 and a pharmaceutically acceptable carrier.

25       14. A pharmaceutical composition comprising the aroylaniline derivative of claim 7 and a pharmaceutically acceptable carrier.

30       15. A method for treating a viral infection of a host which comprises administering to said host an antiviral effective amount of an aroylaniline derivative of claim 1.

16. The method of claim 15, wherein the viral infection is an HIV infection.

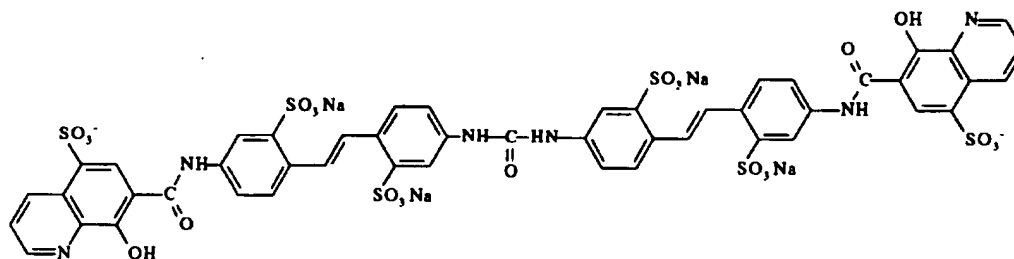
17. The method of claim 16, wherein the aroylaniline derivative is a compound having the formula:



5

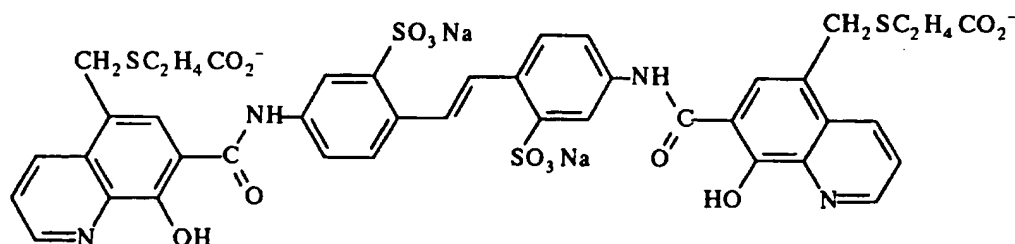
18. The method of claim 16, wherein the aroylaniline derivative is a compound having the formula:

10



15

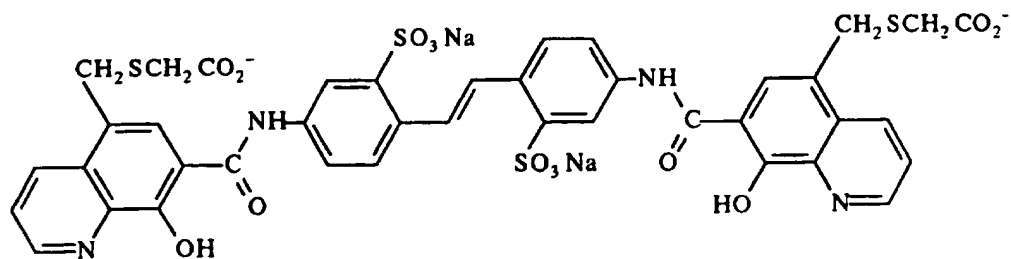
19. The method of claim 16, wherein the aroylaniline derivative is a compound having the formula:



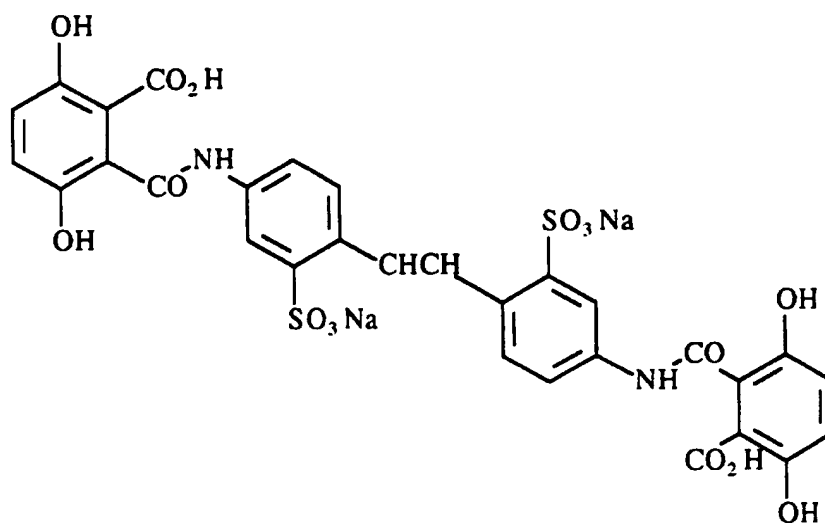
20

20. The method of claim 16, wherein the aroylaniline derivative is a compound having the formula:

54

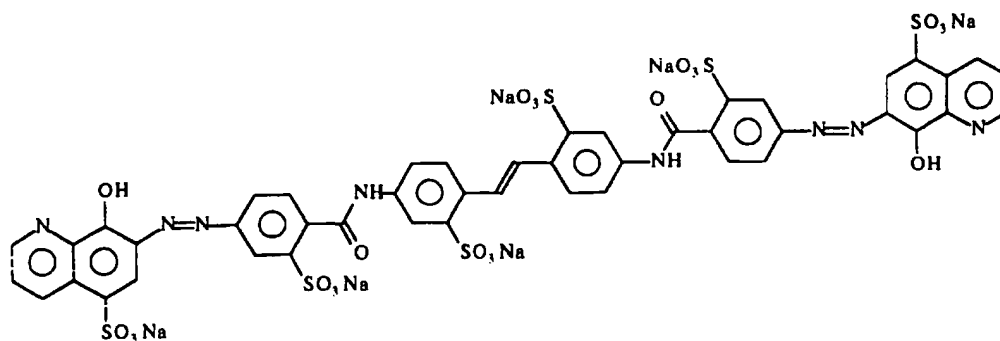


- 5            21. The method of claim 16, wherein the aroylaniline derivative is a compound having the formula:



10

22. The method of claim 16, wherein the aroylaniline derivative is a compound having the formula:



15



55

23. A method for inhibiting replication of a retrovirus comprising contacting said retrovirus with a retrovirally effective amount of an aroylaniline derivative according to claim 1.

5

24. The method of claim 23, wherein said retrovirus is a human immunodeficiency virus.

25. A method for treating a retroviral infection in a mammal comprising administering a retrovirally effective amount of an aroylaniline derivative of claim 1.

10

26. The method of claim 25, wherein said mammal is a human.

15

27. The method of claim 26, wherein said retroviral infection is an HIV infection.

28. A method for preventing a mammal from infection by a virus comprising administering a prophylactically effective amount of an aroylaniline derivative of claim 1.

20

29. The method of claim 28, wherein said virus is a retrovirus.

25

30. The method of claim 29, wherein said retrovirus is a human immunodeficiency virus.

31. The method of claim 28, wherein said mammal is a human.

30

32. A method for inhibiting replication of a virus *in vitro* comprising contacting said virus with an anti-virally effective amount of an aroylaniline derivative of claim 1.

35

56

33. The method of claim 32, wherein said virus is a retrovirus.

34. The method of claim 33, wherein said retrovirus  
5 is a human immunodeficiency virus.

1/3

FIG. 1

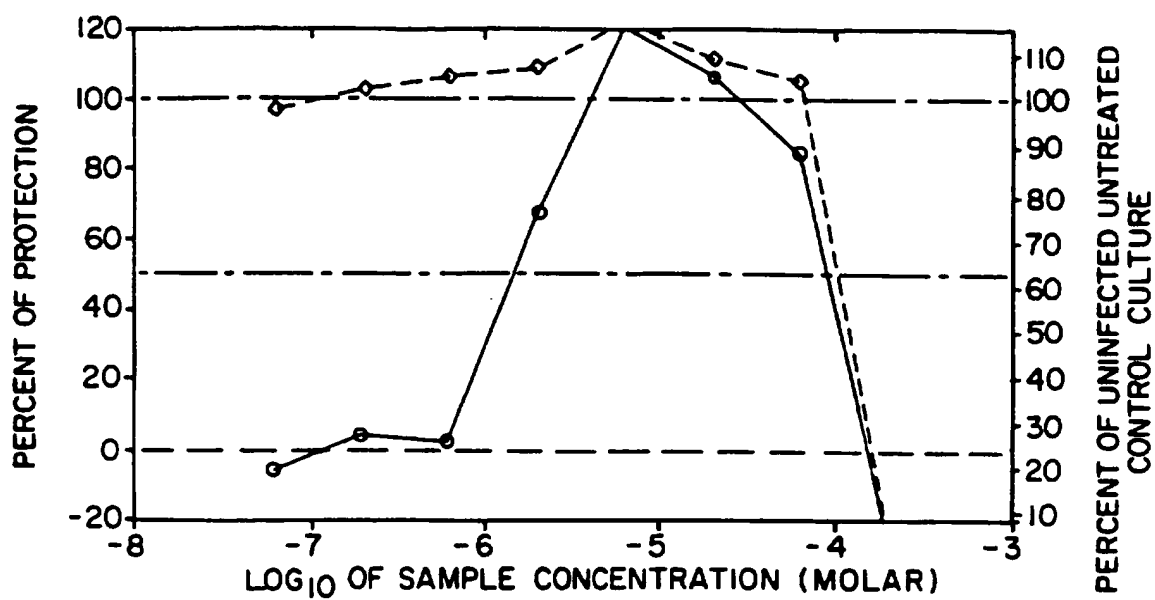
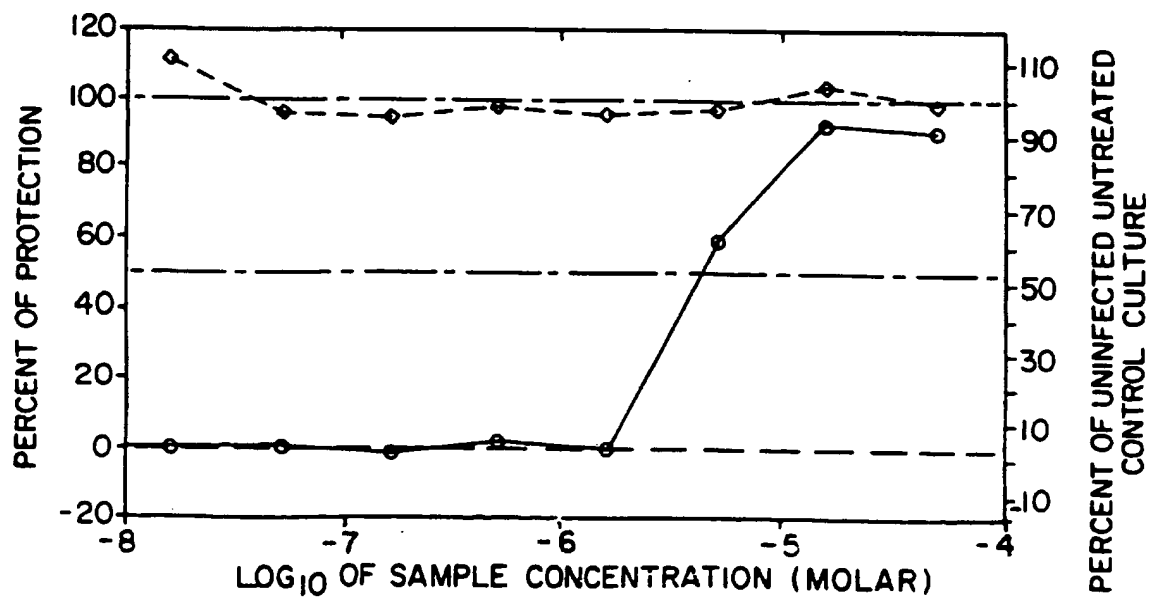


FIG. 2



2 / 3

FIG. 3

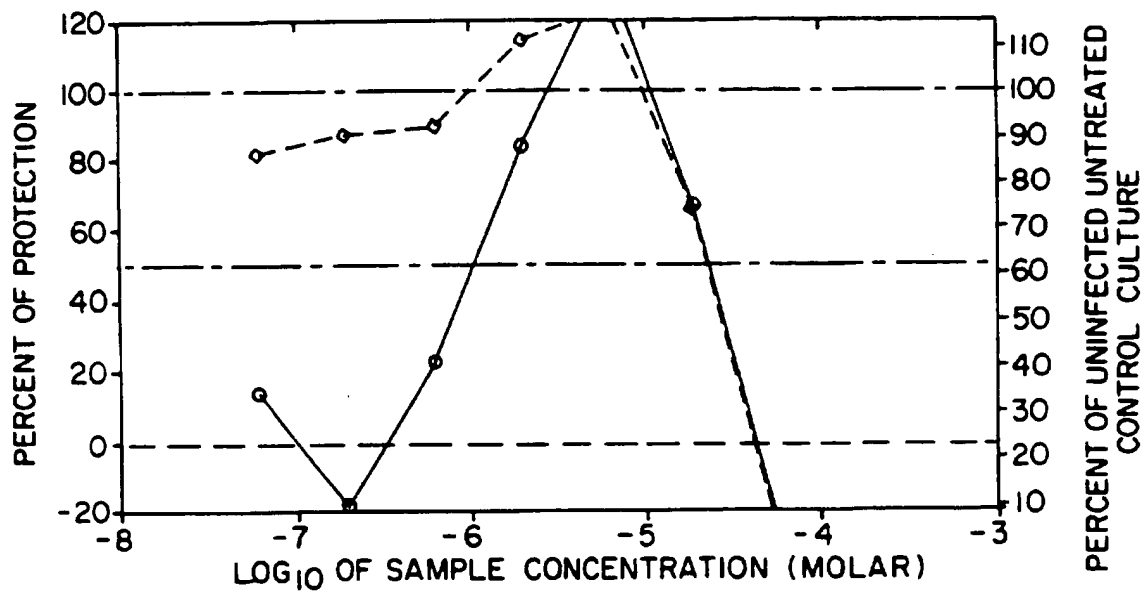
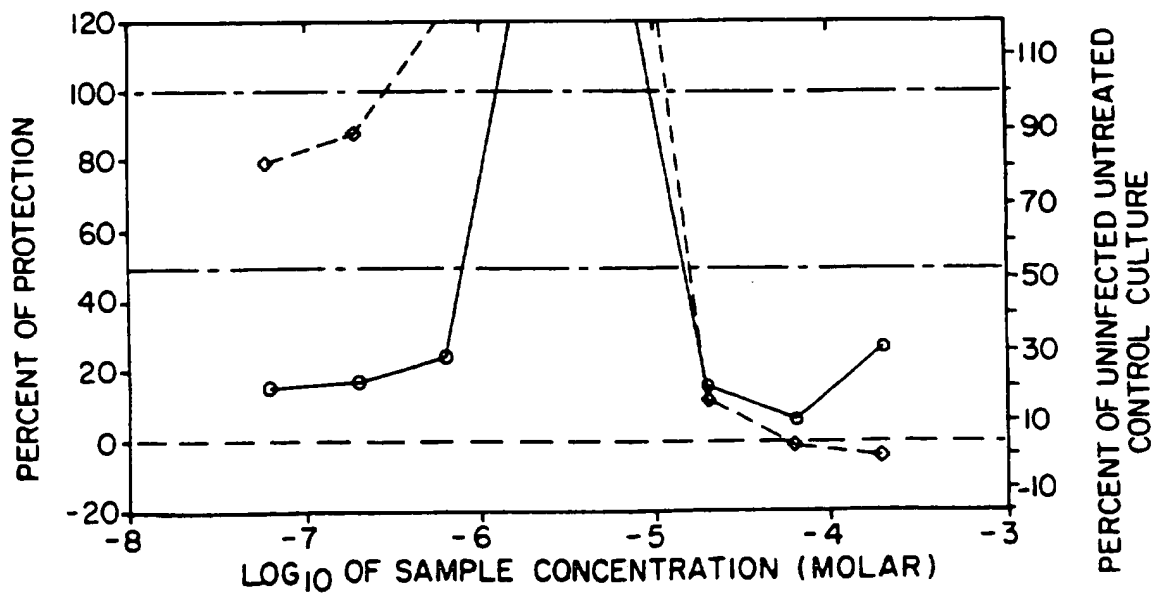


FIG. 4



3/3

FIG. 5

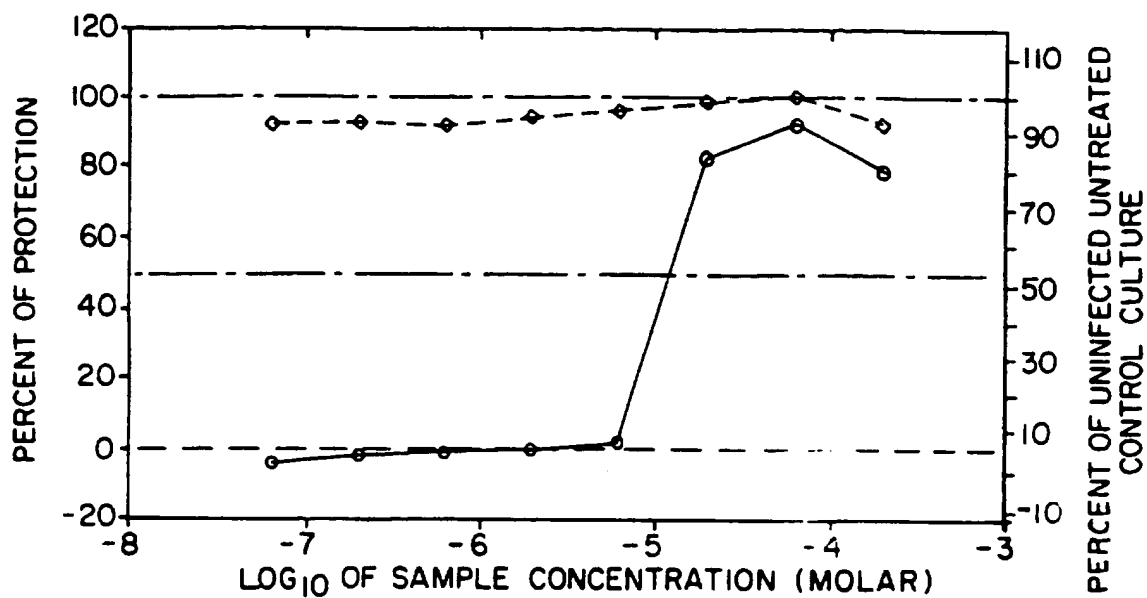
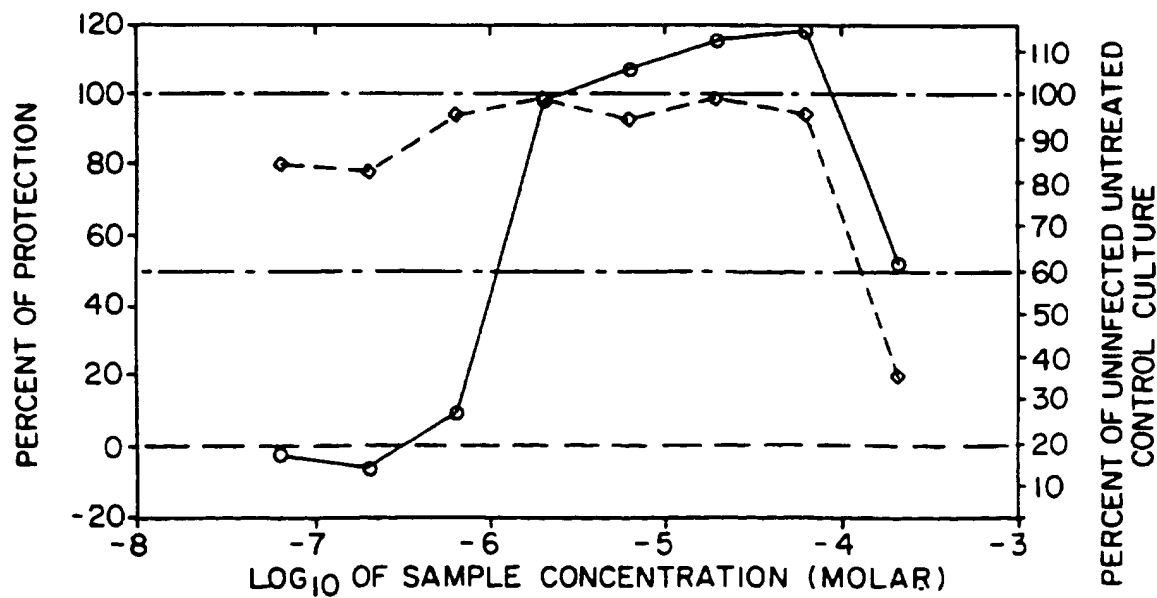


FIG. 6



# INTERNATIONAL SEARCH REPORT

International Application No  
PCT/US 96/02134

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C07D215/48 C07D215/38 C07C309/51 A61K31/47 A61K31/185

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C07D C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	JOURNAL OF THE AMERICAN CHEMICAL SOCIETY, vol. 76, no. 10, 26 May 1954, DC US, pages 2725-2731, XP002005436 W. HEIN ET AL.: "Optical Bleaching Agents. I. Derivatives of Dichloroaminostilbenedisulfonic Acid" see table I  --- -/--	1

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*Z\* document member of the same patent family

Date of the actual completion of the international search

12 June 1996

Date of mailing of the international search report

20.06.96

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+ 31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+ 31-70) 340-3016

Authorized officer

Zervas, B

## INTERNATIONAL SEARCH REPORT

Int. Application No.

PCT/US 96/02134

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	CHEMICAL ABSTRACTS, vol. 85, no. 5, 2 August 1976 Columbus, Ohio, US; abstract no. 32589v, I. V. ALEKSANDROV ET AL.: "Diacyl and diacyldisulfamide derivatives of 4,4'-diaminostilbene-2,2'-disulfonic acid" page 359; column 2; XP002005439 & DEPOSITED PUB., 1972, USSR, pages 4341-4372, see abstract	1
X	--- CHEMICAL ABSTRACTS, vol. 52, no. 11, 10 June 1958 Columbus, Ohio, US; abstract no. 8810h, YASUSHI OHYAMA ET AL.: "Silver halide emulsion" XP002005440 see abstract & JP,A,00 005 919 (MITSUBISHI PAPER MILLS) 19 July 1956	1
X	--- DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005441 see BRN=2928122 & CHEMICAL ABSTRACTS, vol. 72, no. 24, 1970 Columbus, Ohio, US; abstract no. 122893,	1
X	--- DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005442 see BRT=2193992 & KOGYO KAGAKU ZASSHI , vol. 74, no. 1971 , pages 729-732,	1
X	--- DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005443 see BRN=2795085 & YUKI GOSEI KAGAKU KYOKAISHI, vol. 30, 1972, pages 818-819, --- -/--	1

# INTERNATIONAL SEARCH REPORT

Int. Patent Application No  
PCT/US 96/02134

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005444 see BRT=2785594 & JUSTUS LIEBIGS ANN. CHEM, vol. 692, 1966, pages 26-41, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005445 see BRT=2928923 & CHEMICAL ABSTRACTS, vol. 72, no. 24, 1970 Columbus, Ohio, US; abstract no. 122893, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005446 see BRN=2826422 & CHEMICAL ABSTRACTS, vol. 53, no. 21, 10 November 1959 Columbus, Ohio, US; abstract no. 19944b, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005447 see BRN=2797601 & CHEMICAL ABSTRACTS, vol. 72, no. 24, 1970 Columbus, Ohio, US; abstract no. 122893, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005448 see BRN=2928566 & CHEMICAL ABSTRACTS, vol. 72, no. 24, 1970 Columbus, Ohio, US; abstract no. 122893, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005449 see BRN=3192102 & DE,C,250 342 (BAYER & CO.) 1914 ---	1

-/--



## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 96/02134

C(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005450 see BRN=2930702 & J. SOC.DYERS COLOUR., vol. 81, 1965, pages 102-104, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005451 see BRN=4050045 & BIOCHEMISTRY, vol. 18, 1979, pages 4505-4510, ---	1
X	DATABASE CROSSFIRE Beilstein Informationssysteme GmbH, Frankfurt DE XP002005452 see BRN=2932158 & YUKI GOSEI KAGAKU KYOKAISHI, vol. 29, 1971, page 519-522 ---	1
A	WO,A,93 16992 (MERRELL DOW PHARMACEUTICALS) 2 September 1993 see claims; examples ---	1-34
A	WO,A,94 14763 (PROCEPT) 7 July 1994 see claims; examples ---	1-34
A	JOURNAL OF MEDICINAL CHEMISTRY, vol. 36, no. 14, 9 July 1993, WASHINGTON US, pages 1996-2003, XP002005437 PREM MOHAN ET AL. : "Structure-Activity Relationship Studies with Symmetric Naphthalenesulfonic Acid Derivatives. Synthesis and Influence of Spacer and Naphthalenesulfonic Acid Moiety on Anti-HIV-1 Activity" cited in the application see the whole document ---	1-34
	---	

-/--

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 96/02134

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>JOURNAL OF MEDICINAL CHEMISTRY,  vol. 35, no. 26, 25 December 1992,  WASHINGTON US,  pages 4846-4853, XP002005438  GHEE T. TAN ET AL.: "Potential Anti-AIDS  Naphthalenesulfonic Acid Derivatives.  Synthesis and Inhibition of HIV-1 Induced  Cytopathogenesis and HIV-1 and HIV-2  Reverse Transcriptase Activities"  cited in the application  see the whole document  -----</p>	1-34

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 96/02134

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:  
Although claims 15 to 34 are directed to a method of treatment of the human/animal body, the search has been carried out and based on the alleged effects of the compound/composition.
2. ☐ Claims Nos.:  
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:  
  
Claims 1,8,15,16 23-34 have been searched incompletely.  
Please see attached sheet ./.
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.

# INTERNATIONAL SEARCH REPORT

International Application No

PL /US 96/02134

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO-A-9316992	02-09-93	AU-B- 663925	26-10-95
		AU-B- 3481893	13-09-93
		CA-A- 2130360	21-08-93
		EP-A- 0626945	07-12-94
		JP-T- 7504190	11-05-95
		US-A- 5359131	25-10-94
		ZA-A- 9301035	17-09-93
-----			
WO-A-9414763	07-07-94	AU-B- 5850894	19-07-94
-----			